

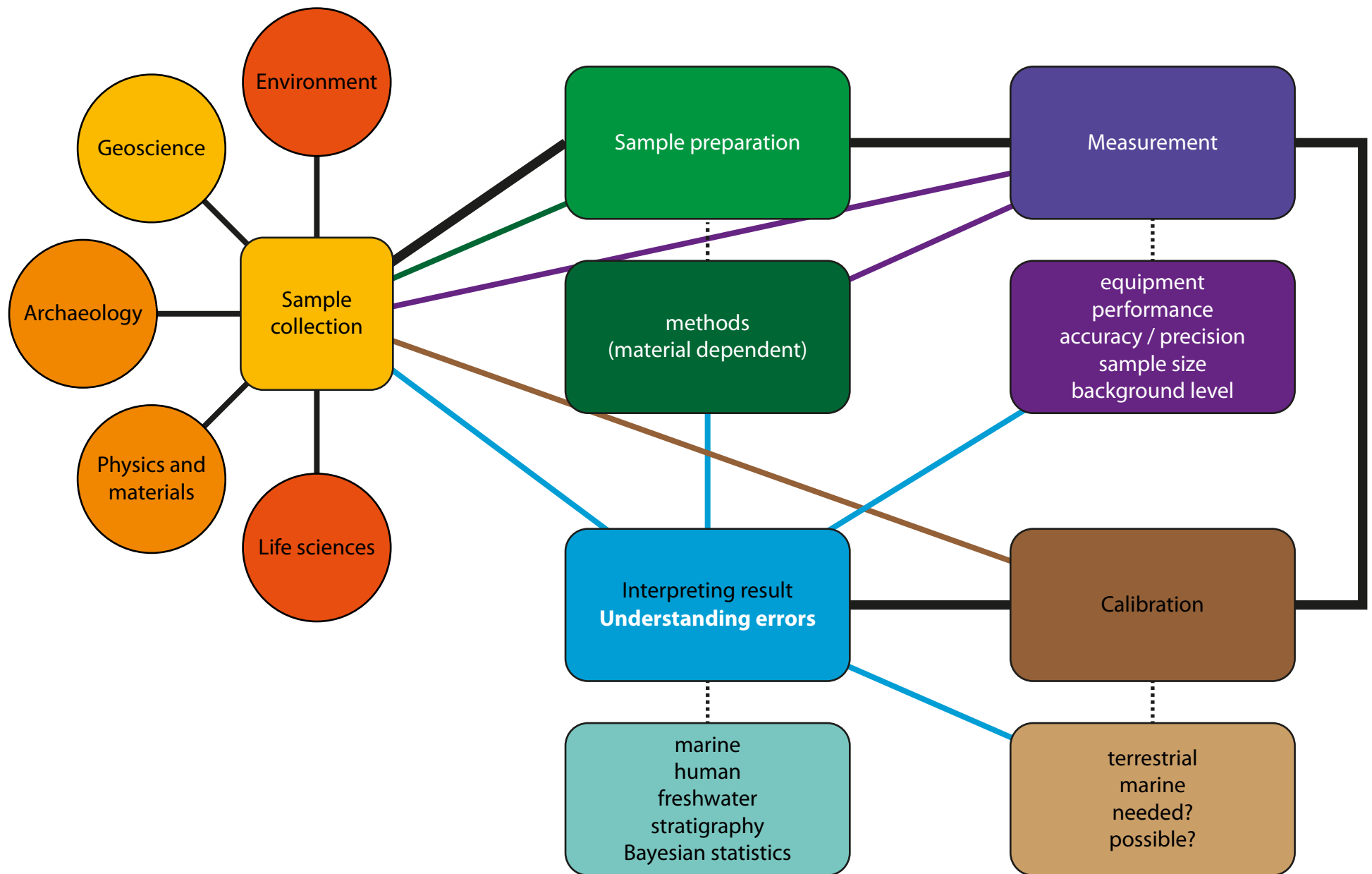
Radiocarbon



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Basic principle - overview



Production and cycling of radiocarbon

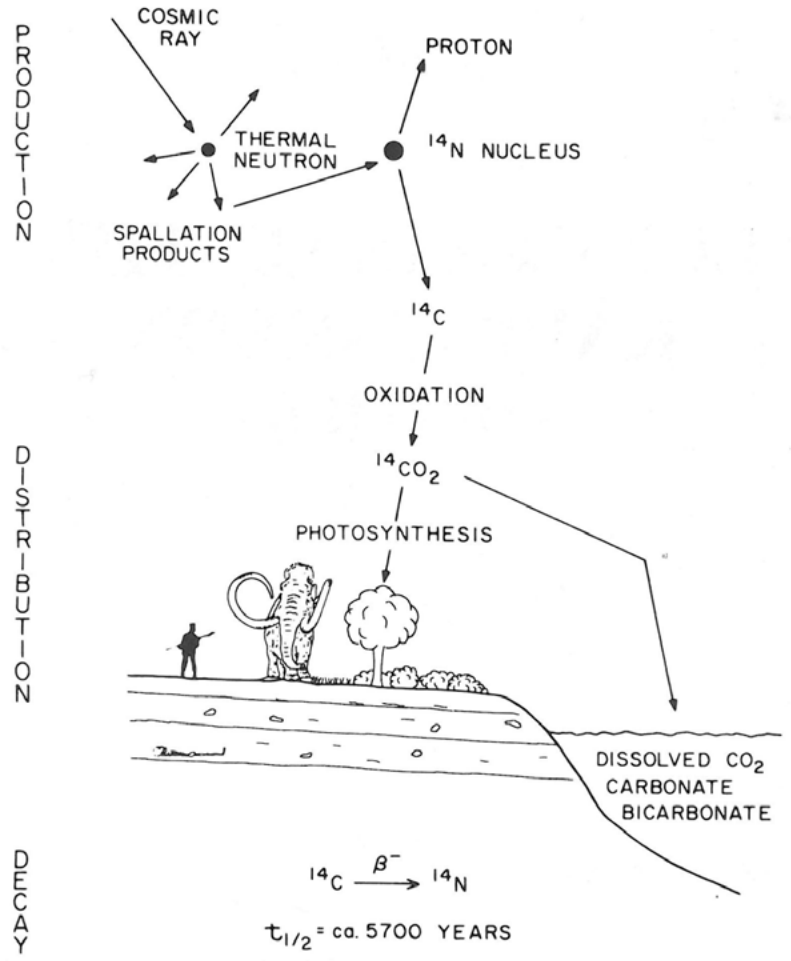
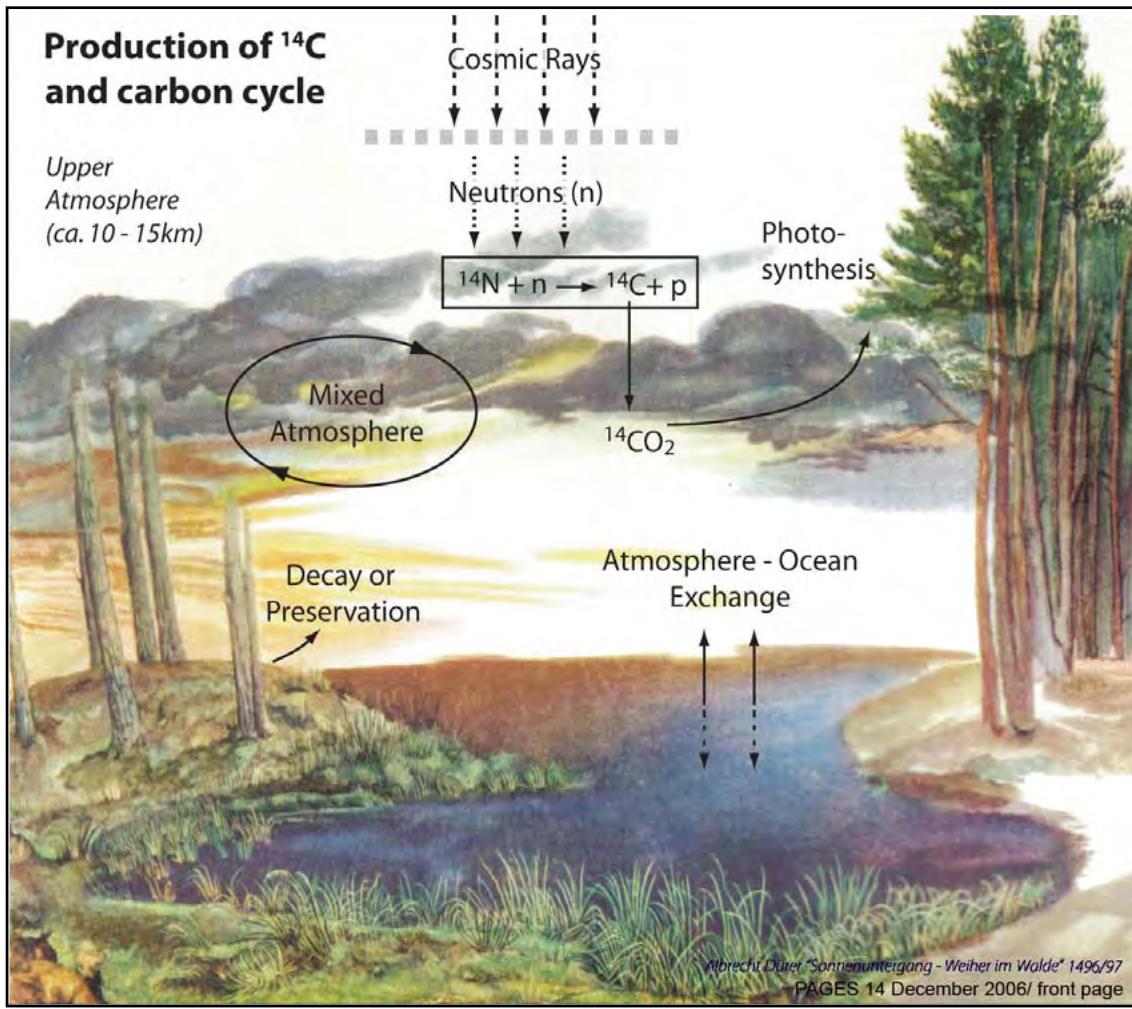
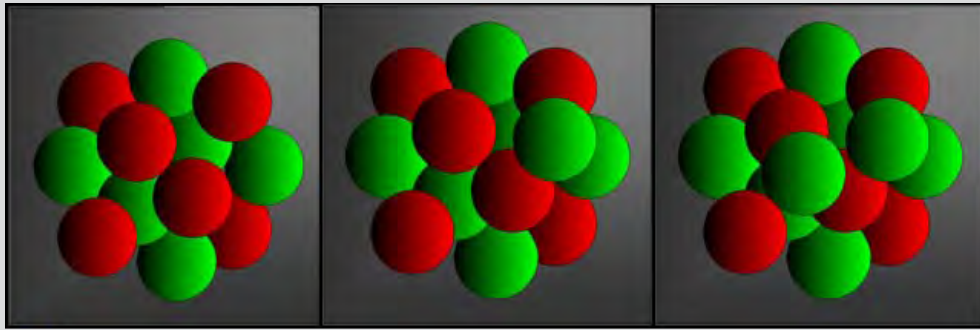


Figure 1.1 Basis of radiocarbon method: production, distribution and decay of ^{14}C . [After Stuiver & Reimer (1985b).]

Basic technique / Dating principle



6 protons
6 neutrons

6 protons
7 neutrons

6 protons
8 neutrons

Libby: Nobel prize (1960) "for his method to use carbon-14 for age determination in archaeology, geology, geophysics and other branches of science"

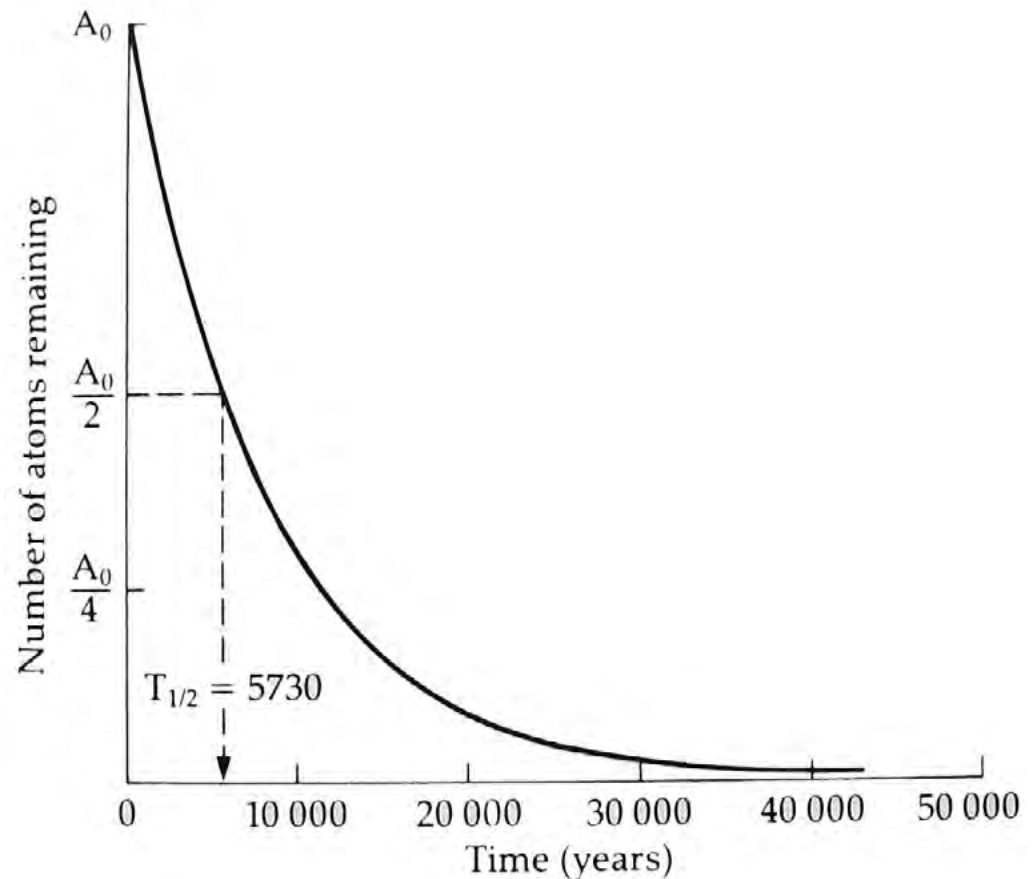


Radioactive decay:

$$A = A_0 \cdot e^{-\frac{t}{\tau}}$$

$$t = -\tau \cdot \ln\left(\frac{A}{A_0}\right)$$

Measure $^{14}\text{C}/^{12}\text{C}$ (if $A_0 \approx \text{constant}$)



Basic technique / Dating principle

Radioactive decay:

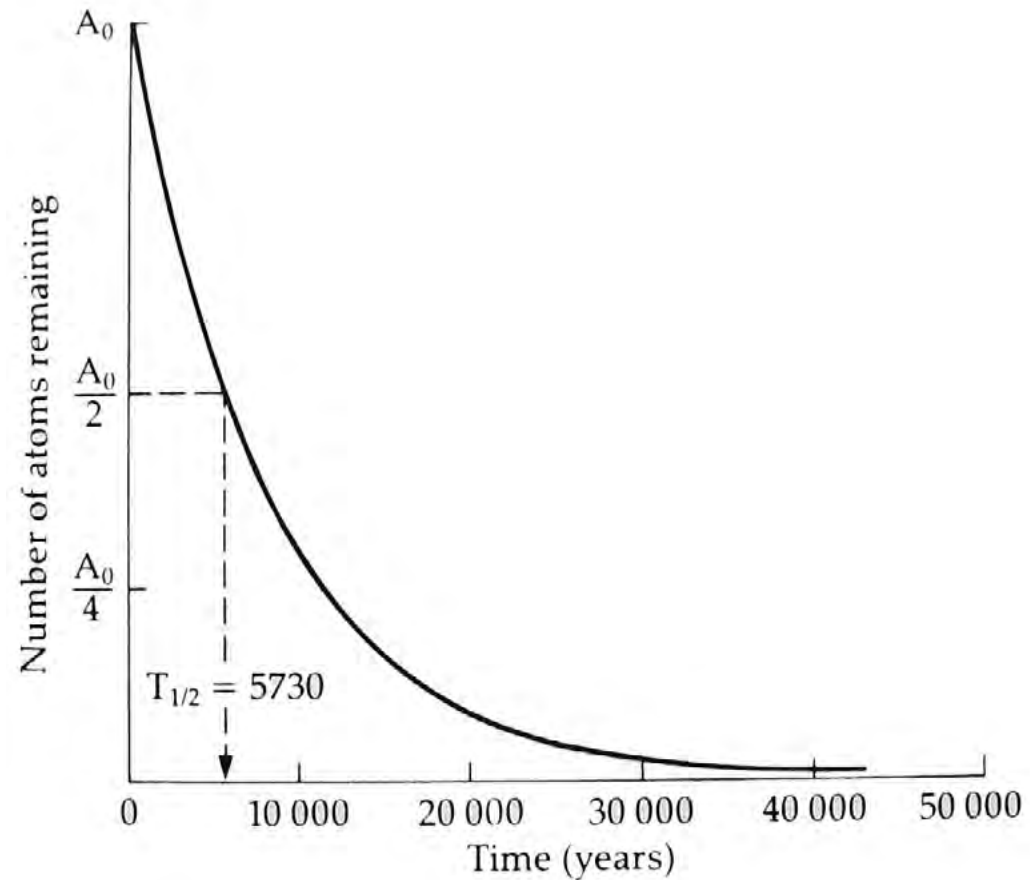
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Measure $^{14}\text{C}/^{12}\text{C}$ (if $A_0 \approx \text{constant}$)

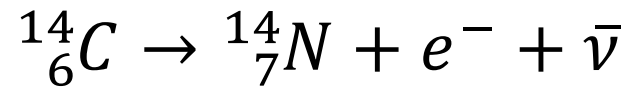
Assumptions:

- ^{14}C activity in each part of the global carbon reservoir (atmosphere, ocean, biosphere) has remained constant over time
- Complete and rapid mixing of ^{14}C
- no alteration of ^{12}C , ^{13}C , ^{14}C since death
- half life is accurately known



Målinger...





Detecting the beta decay
 gas proportional counting
 liquid scintillation

Counting atoms using accelerator
 mass spectrometry (AMS)

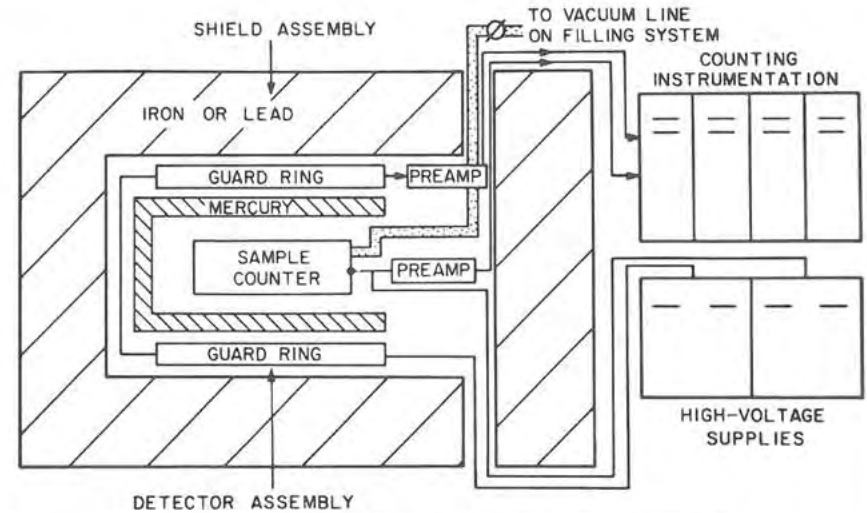


Figure 4.4 Components of a gas decay counting system.

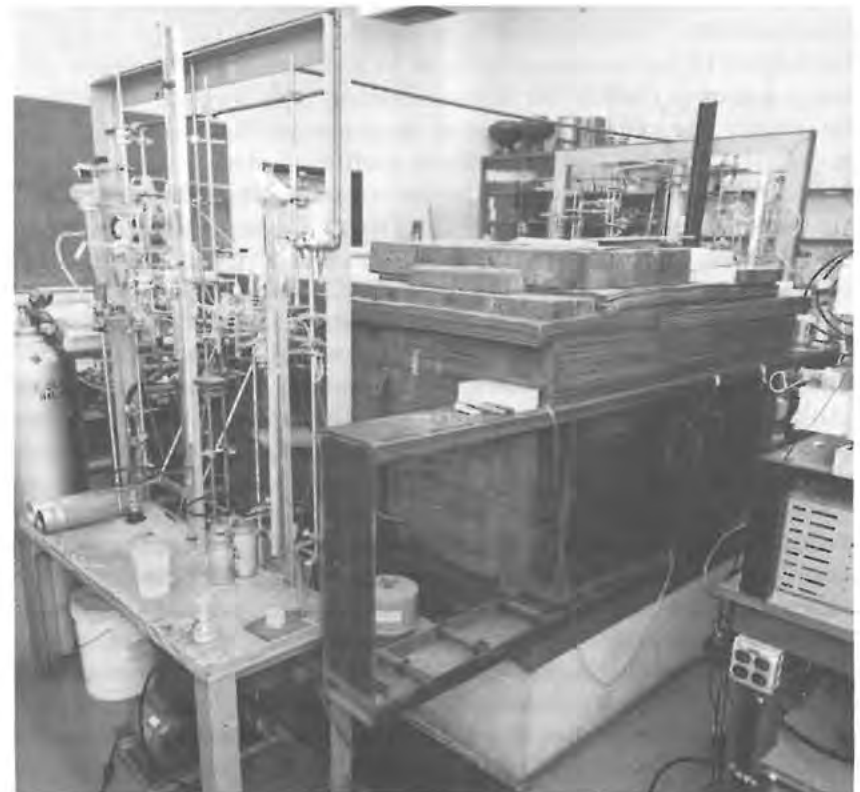
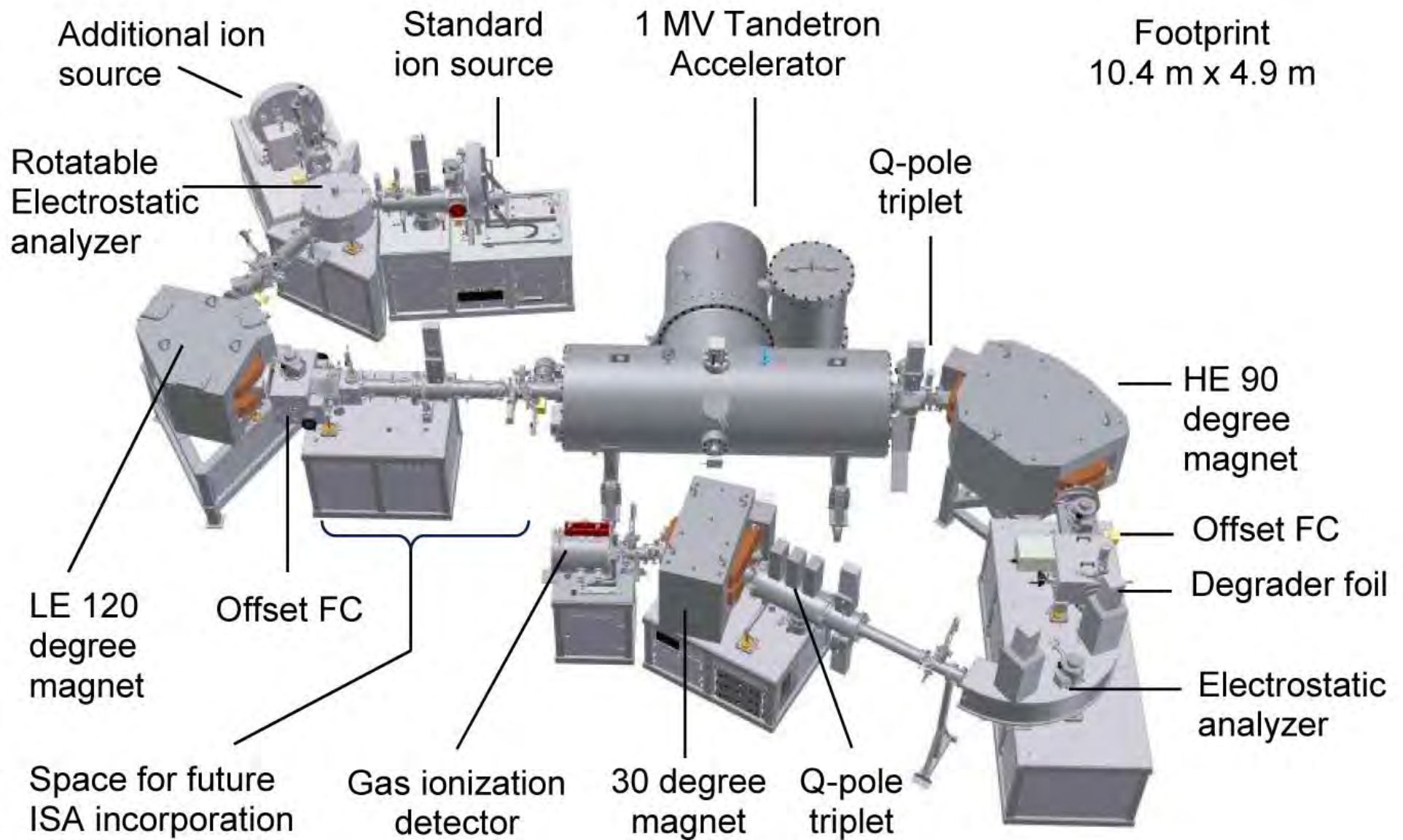


Figure 4.5 Shield assembly for a gas counting system. (Photograph courtesy of H. E. Suess and L. D. Ford, University of California, San Diego.)

^{14}C dating using AMS:

Small samples	(0.02) 0.2 – 1 mg (research!)
Capacity	>1000 pr. year
Measuring time	0.5 – 2 hours
^{14}C precision	0.3% or ± 25 years (research!)
Range	40 – 50.000 years





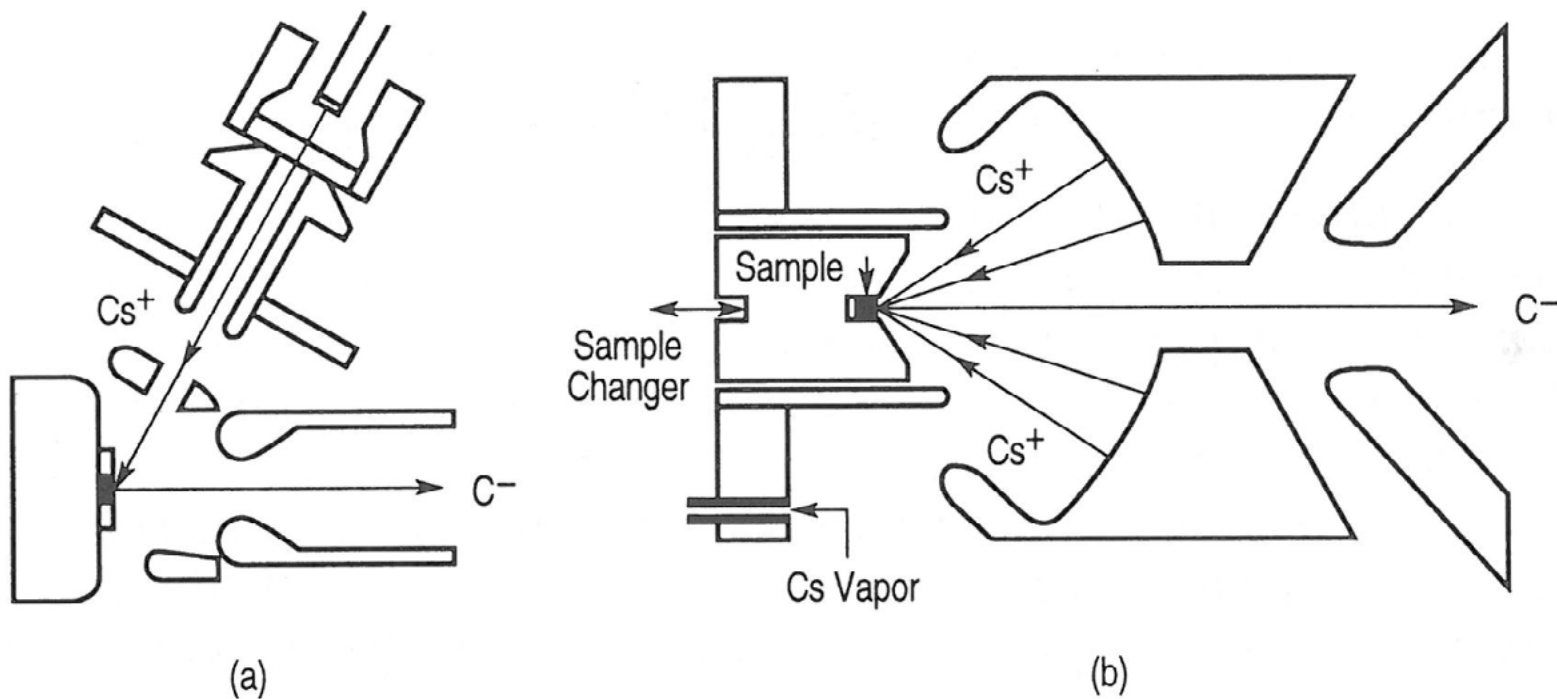
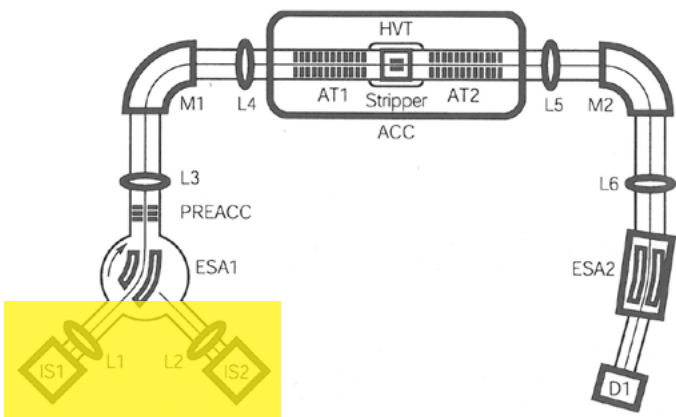


Figure 3.2 Ion source geometries: (a) focused Cs source, and (b) high-intensity source.



Mass	Molecule
12	^{12}C
13	^{12}CH , ^{13}C
14	$^{12}\text{CH}_2$, ^{13}CH , ^{14}C

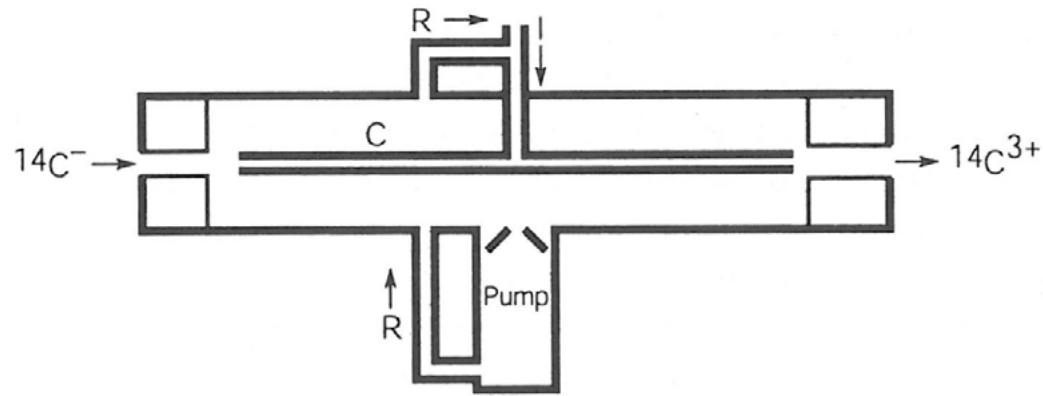
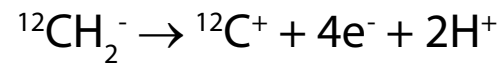
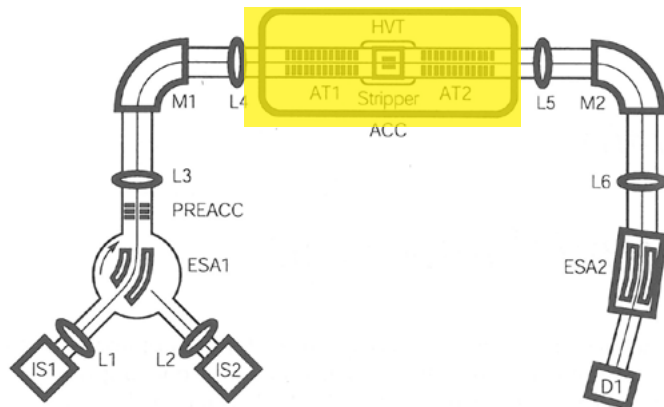
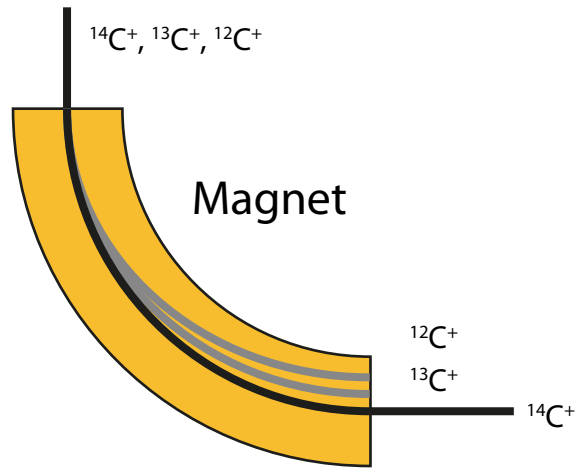


Figure 3.4 Schematic illustration of a gas stripper system in the terminal of a tandem accelerator with recycling of the stripper gas. The incoming negative ion (shown for example as $^{14}\text{C}^-$) is stripped of four electrons as it passes through the gas canal (C), emerging at the exit as a $^{14}\text{C}^{3+}$ ion for further acceleration. The gas is fed into the canal by a controlled needle valve, removed by a turbo pump, and recirculated (R) back into the stripper canal. (From Bonani, G. et al., *Nucl. Instrum. Methods Phys. Res.*, B52, 338, 1990. With permission.)



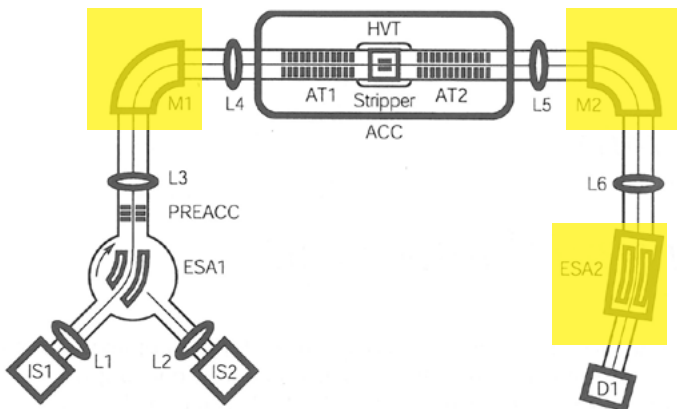


Mass filter:

$$F = m \cdot a$$

$$m \frac{v^2}{r} = q \cdot B \cdot v$$

$$B \cdot r \propto \frac{\sqrt{m \cdot E}}{q}$$



Energy filter:

$$F = m \cdot a$$

$$m \frac{v^2}{r} = q \cdot \epsilon = q \frac{V}{d}$$

$$\frac{Vr}{2d} \propto \frac{E}{q}$$

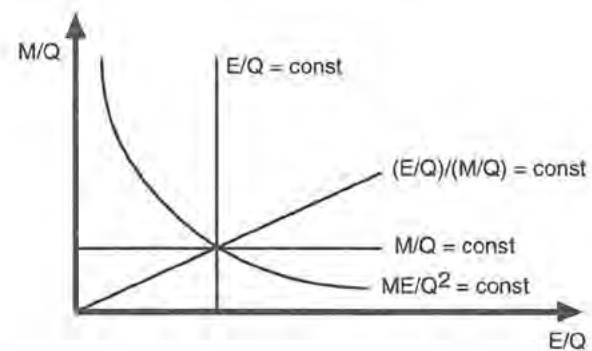
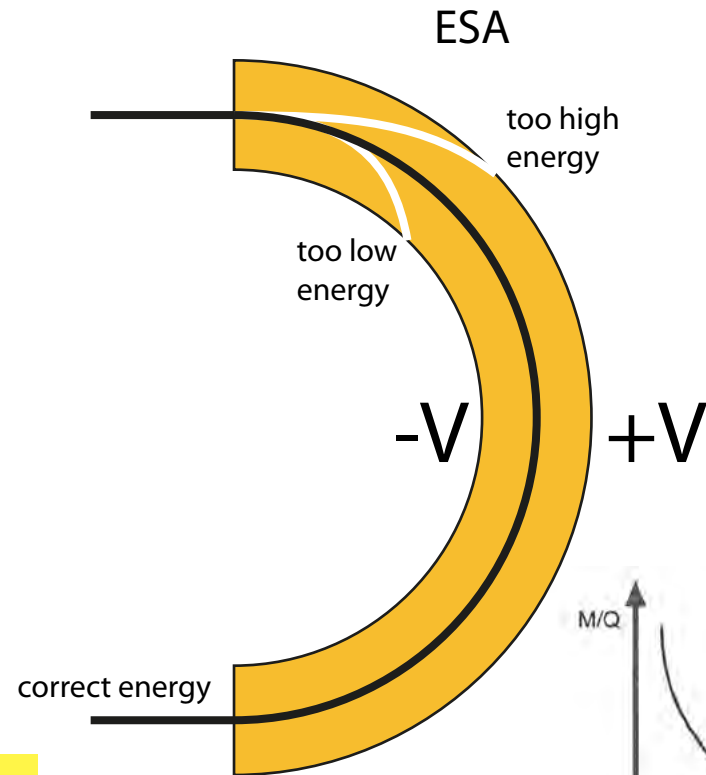


Figure 3.5 Loci of values of M/Q vs. E/Q determined by different beam line analyzers. Ions with equal E/Q ratios will be transmitted identically by an electrostatic analyzer, those with equal ME/Q^2 by a magnetic analyzer, and those with equal (E/Q) and (M/Q) by a velocity (or Wien) filter. Combinations of analyzers can be used to dramatically reduce background events. In order for an ion to pass all analyzers, it must have values of E , M , and Q which are determined by the intersection point of all analyzer loci.

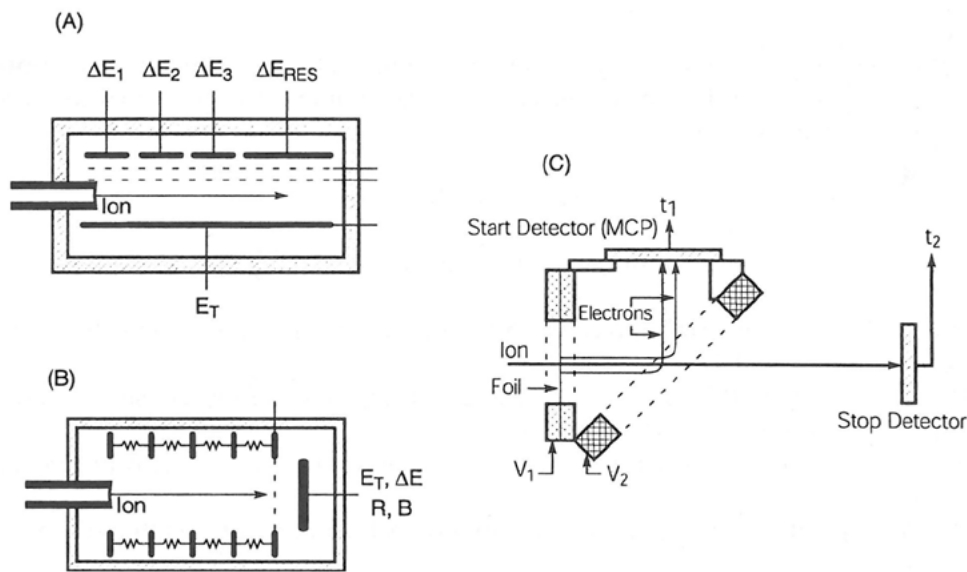


Figure 3.7 Schematic drawings of ion detectors used in AMS: (A) multi-anode gas ionization detector (see Fifield et al., 1987, and Knies and Elmore, 1994b, for details); (B) a Bragg detector; and (C) start and stop detector setup used to determine the time-of-flight ($t_2 - t_1$) for each ion. Dashed lines

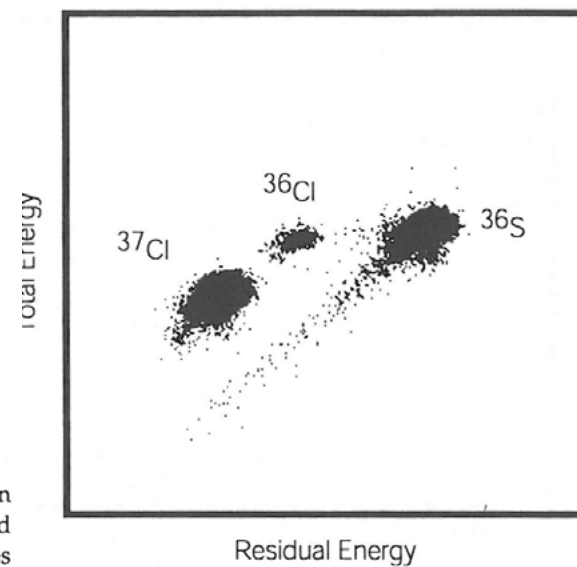
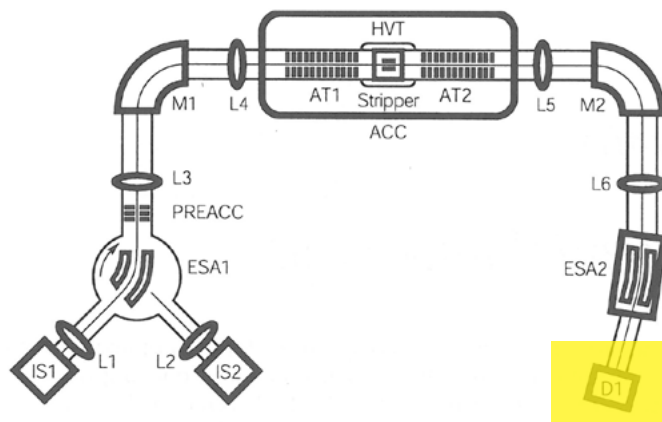


Figure 3.8 Identification of ^{37}Cl , ^{36}Cl , and ^{36}S ions at 130 MeV using the residual energy (last anode plate) and total energy parameters measured with a gas ionization detector. Correlated energy degraded events associated with the ^{37}Cl and ^{36}S peaks (at $<0.1\%$ of peak intensity) are observed. (From Ophel, T.R. et al., *Nucl. Instrum. Methods Phys. Res.*, A272, 734, 1988. With permission.)



Calibration

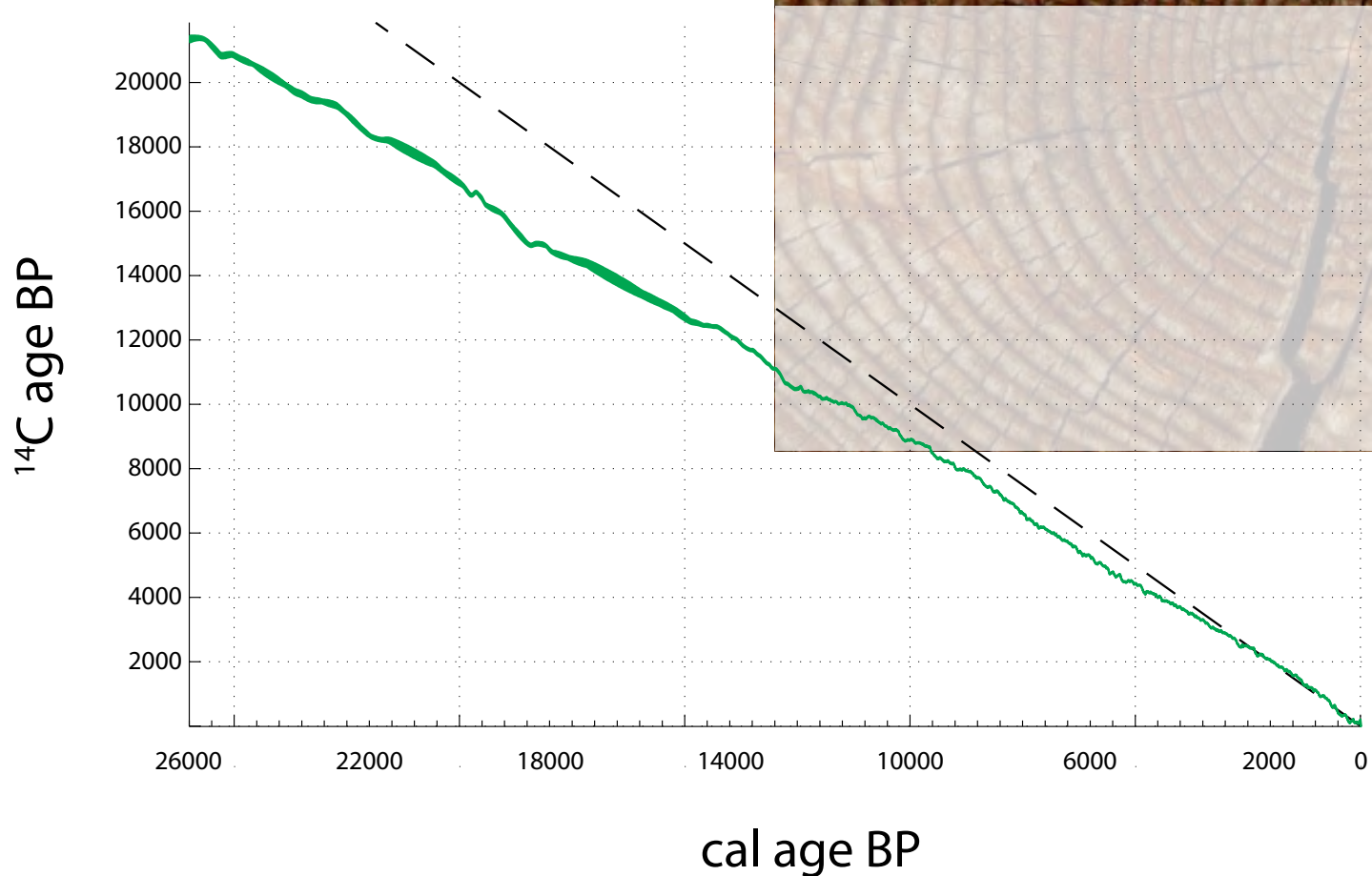


from ^{14}C years to calendar years

Radiocarbon chronology

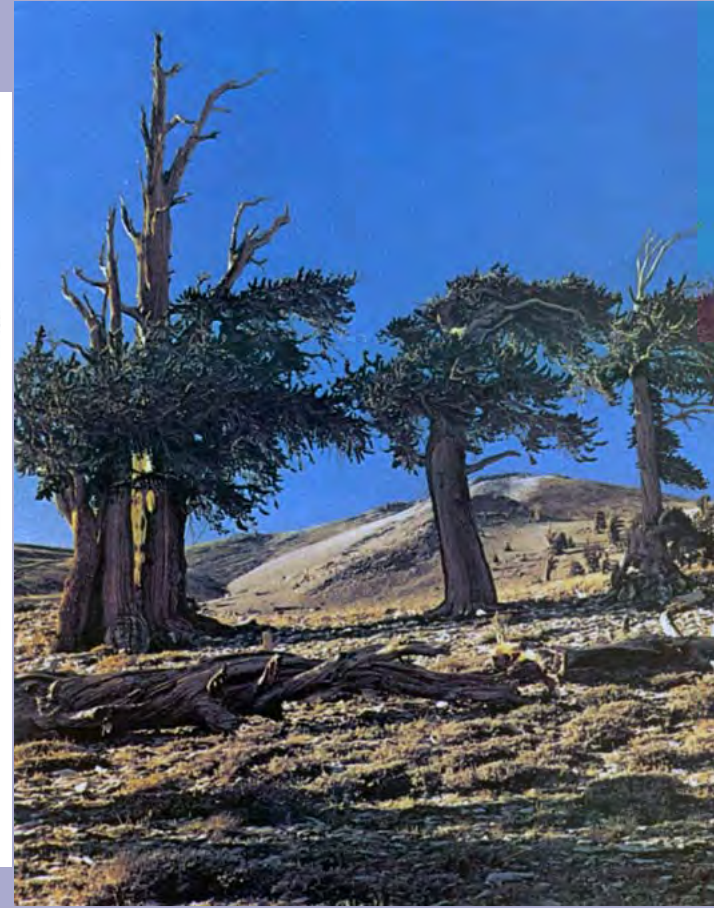
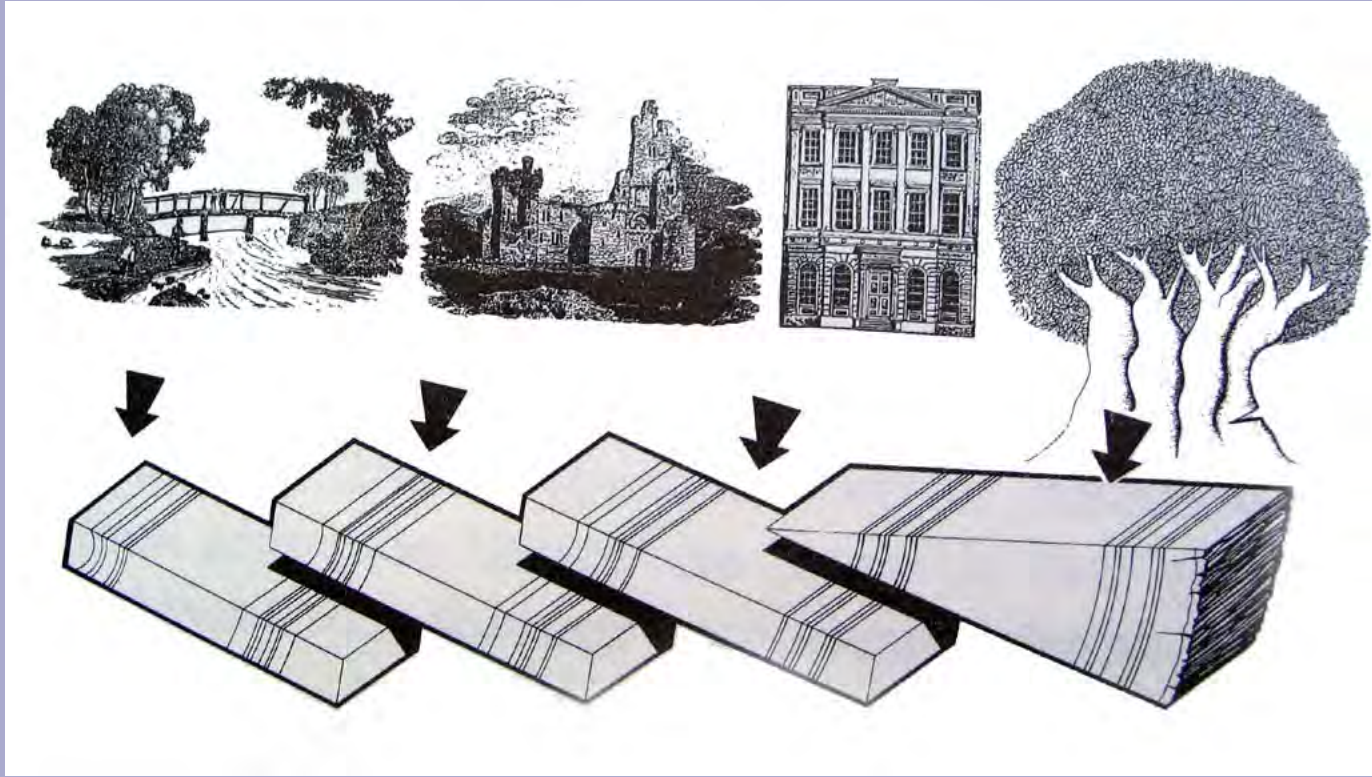
Dendro-chronology

Past atmospheric ^{14}C levels

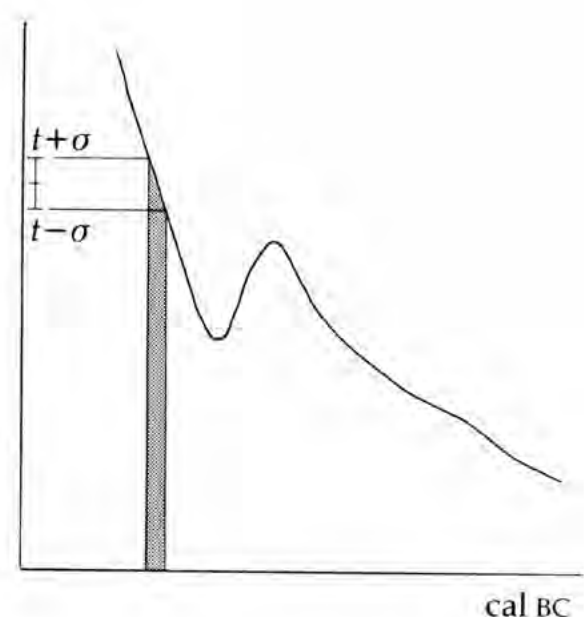
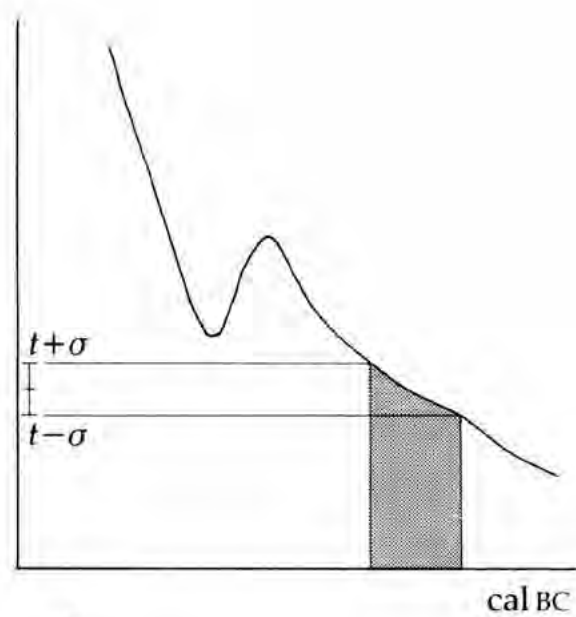
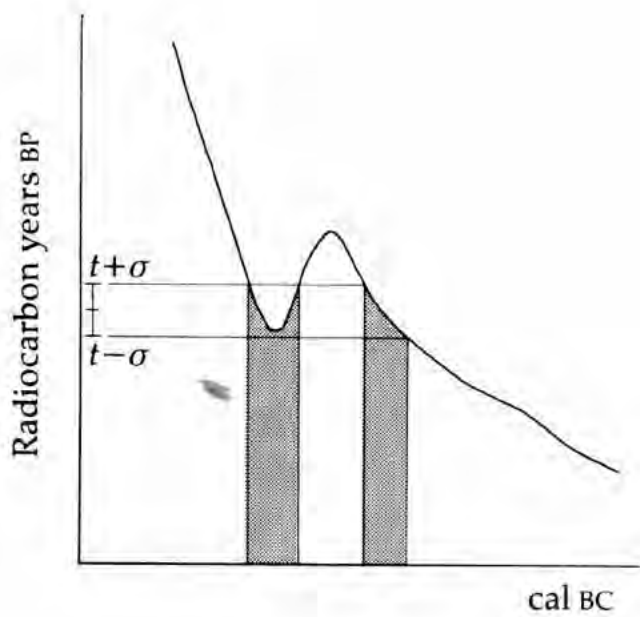


Calibration:
Sample must be in carbon equilibrium with atmosphere

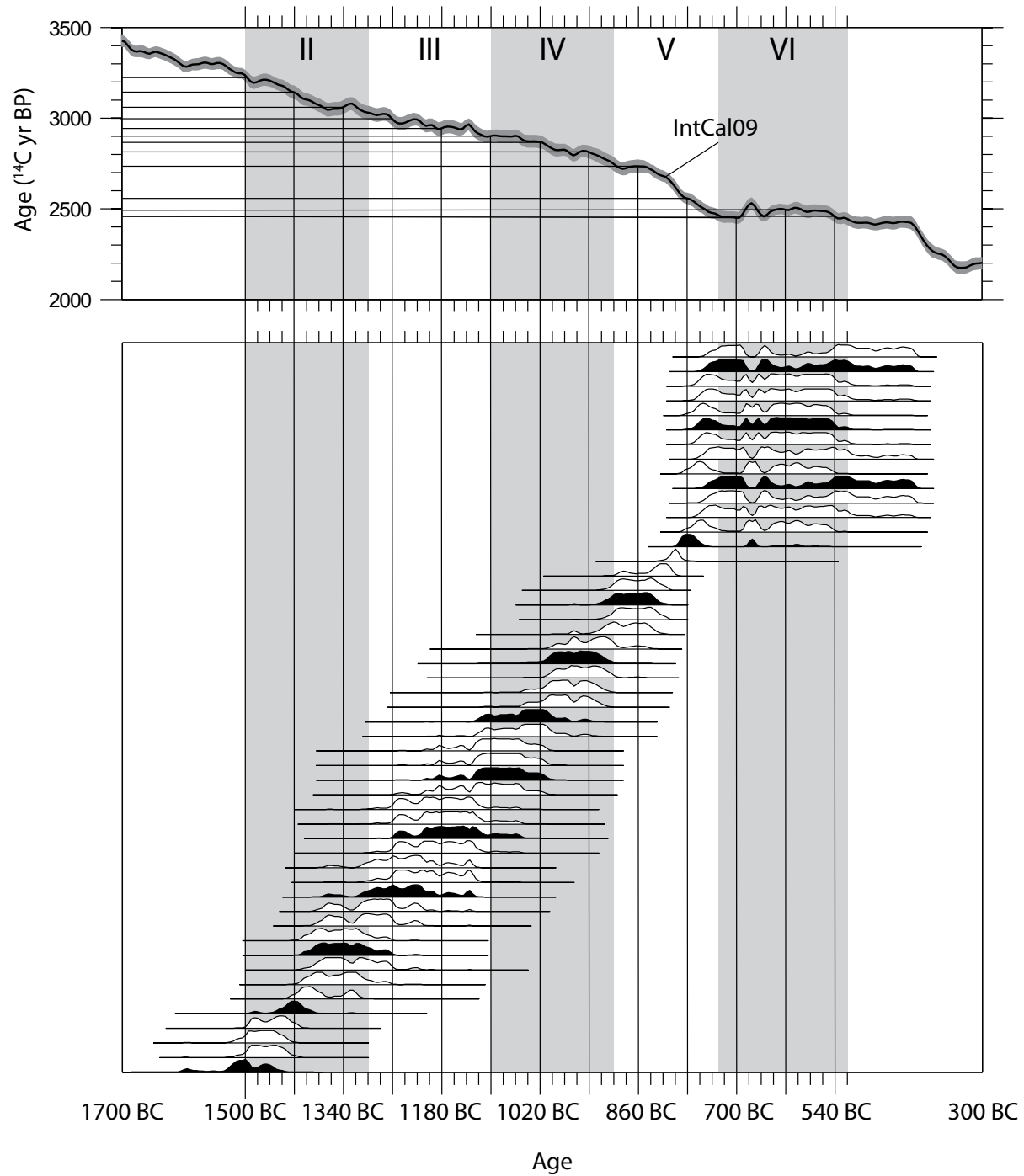
The radiocarbon calibration curve

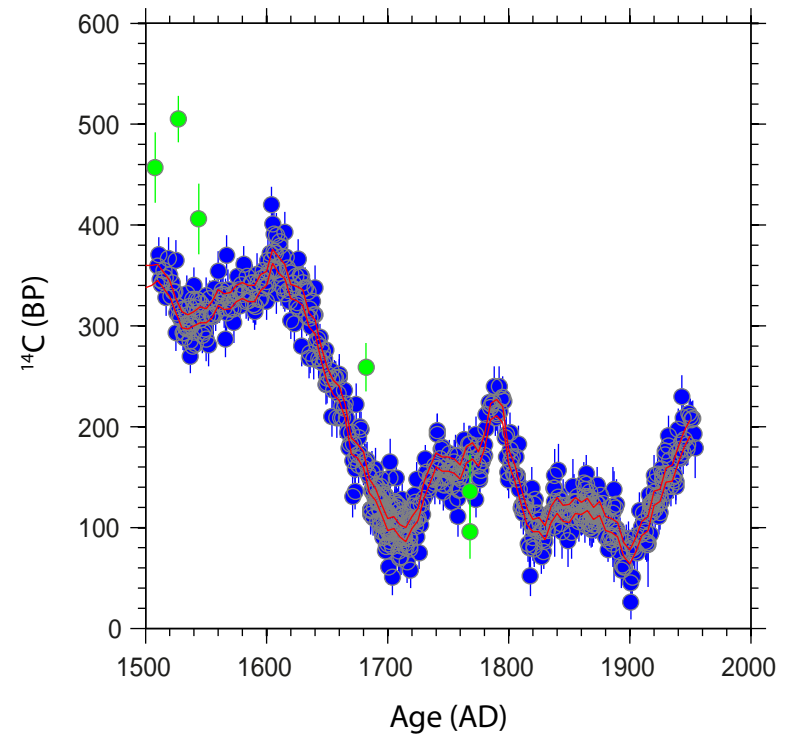
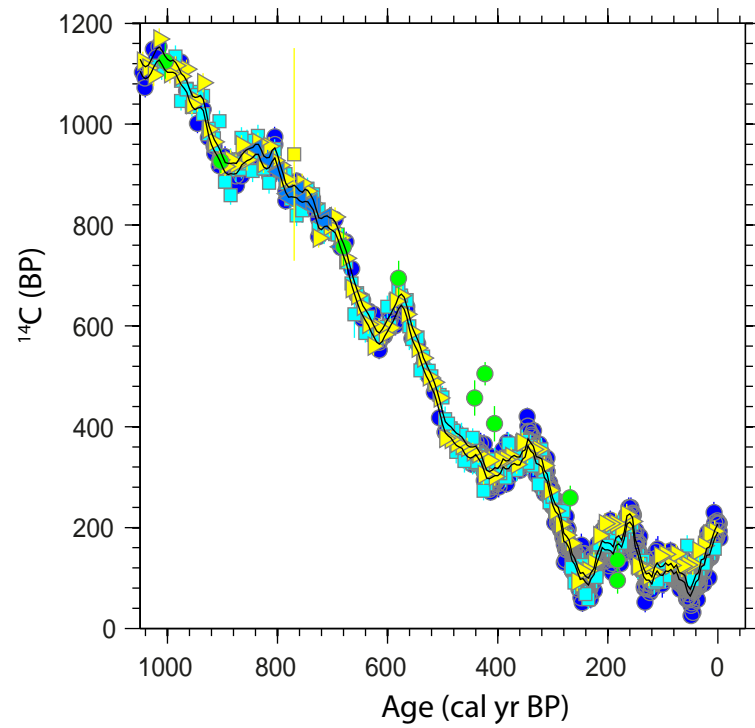


How to convert a ^{14}C date into a calendar date



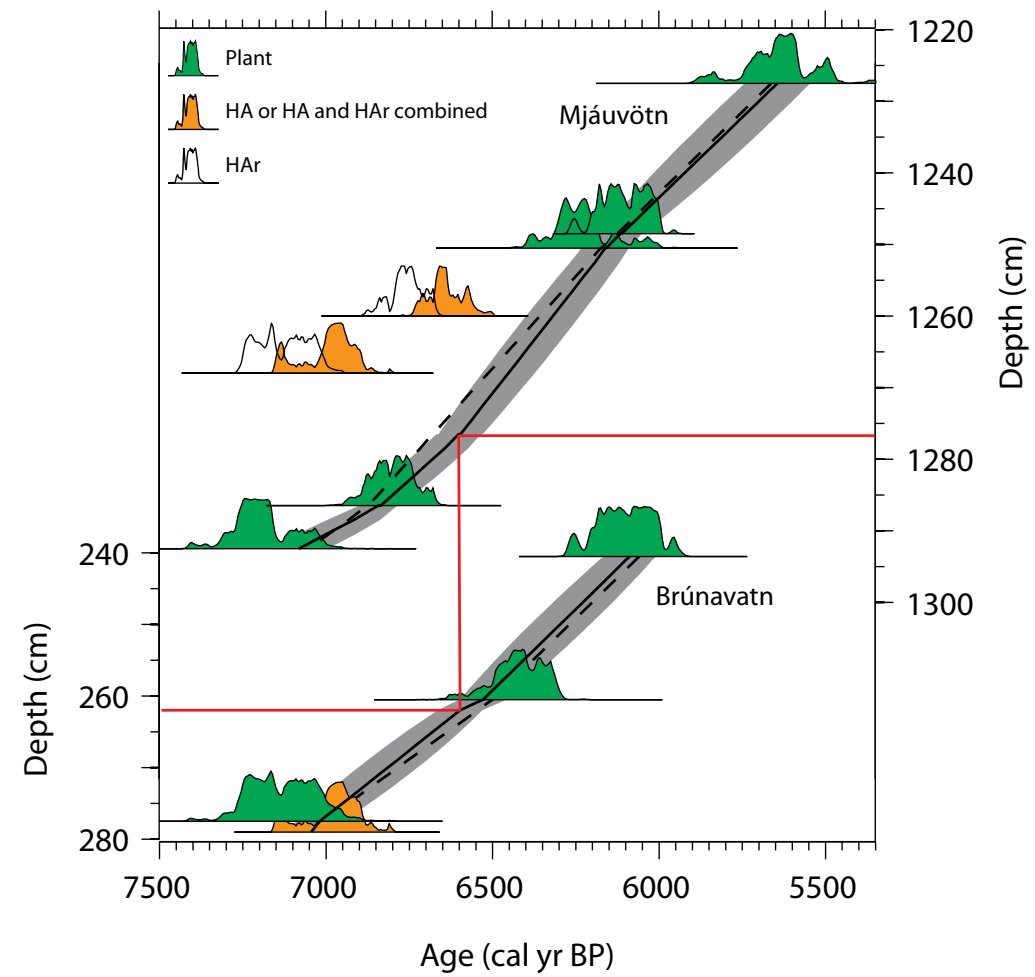
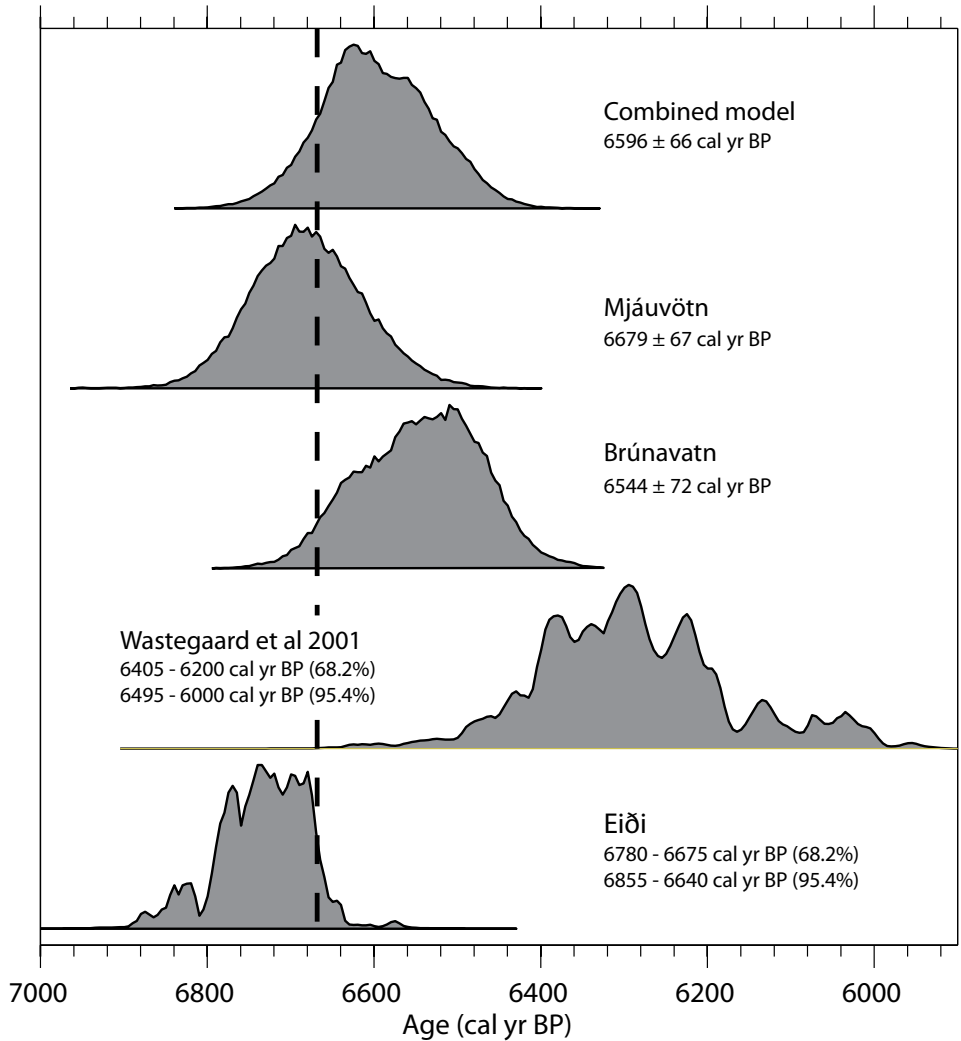
Radiocarbon calibration





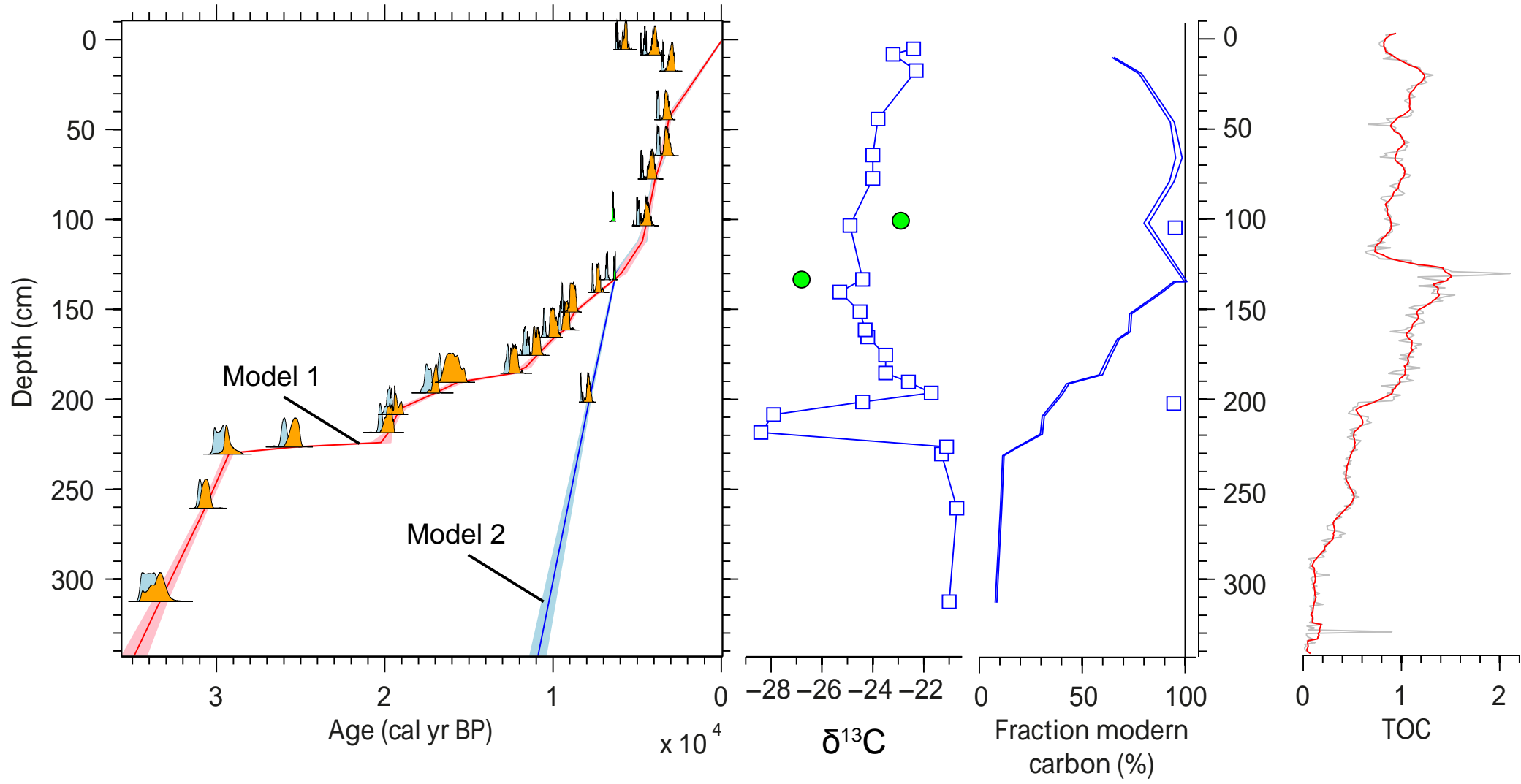
Suess effect - annual data (can the 11 yr solar cycle be observed?)

Linked age models

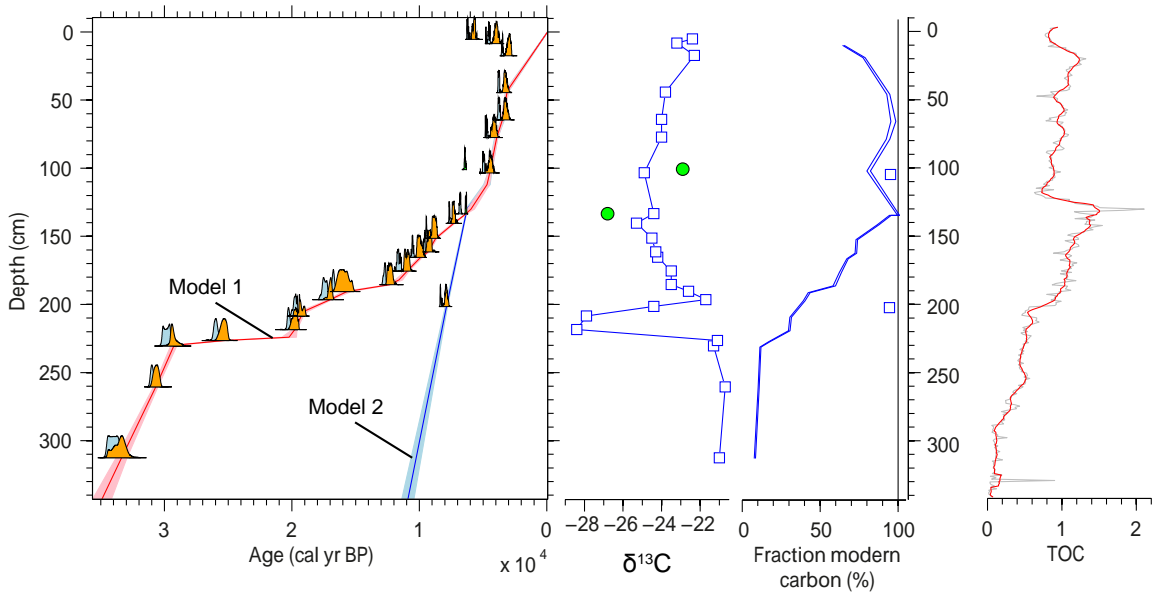




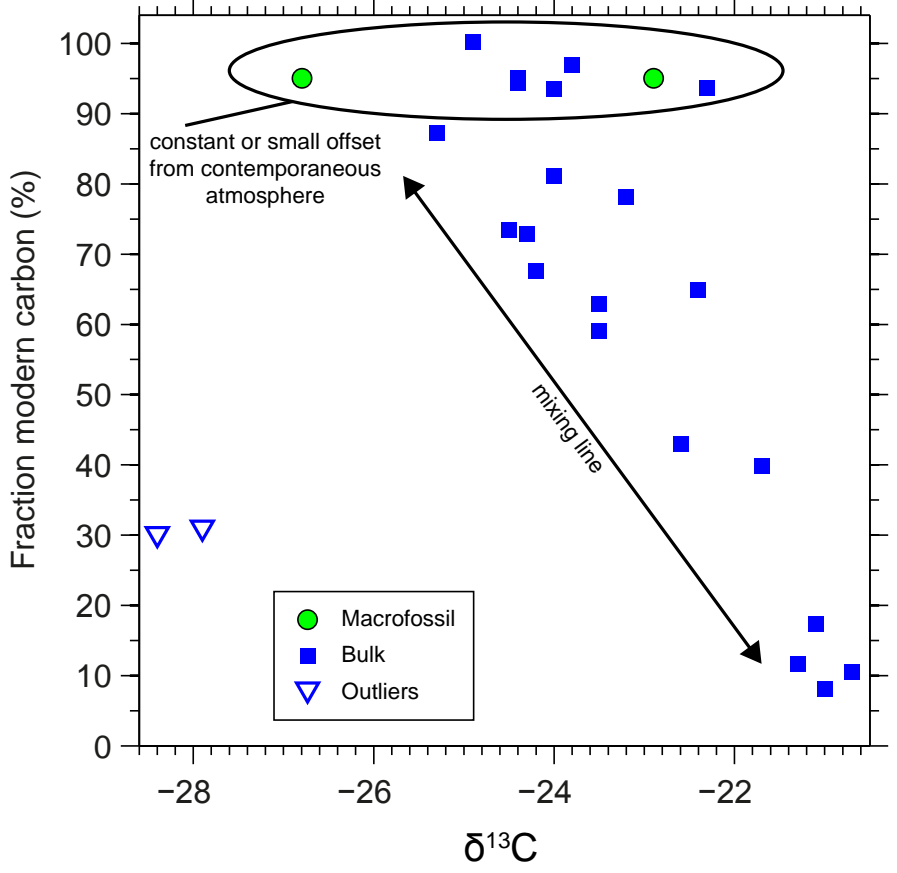
A complicated chronology - but can we provide explanations?



Mixing between two reservoirs



local glacier activity

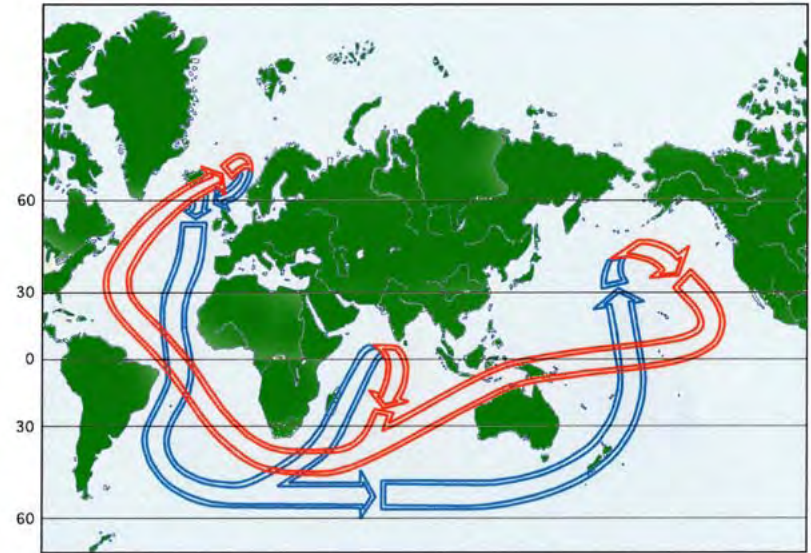
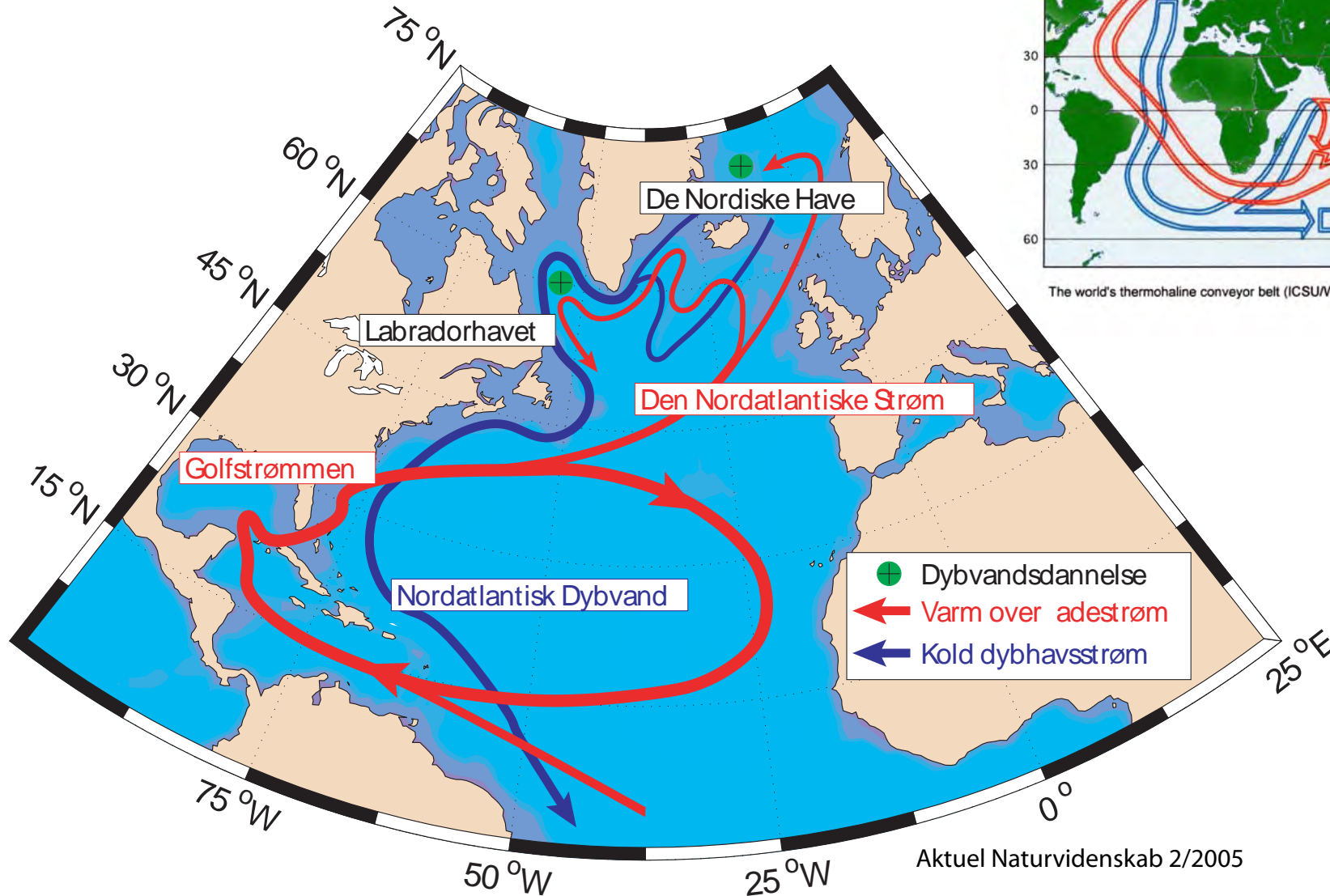


Radiocarbon and the ocean....



Ocean circulation

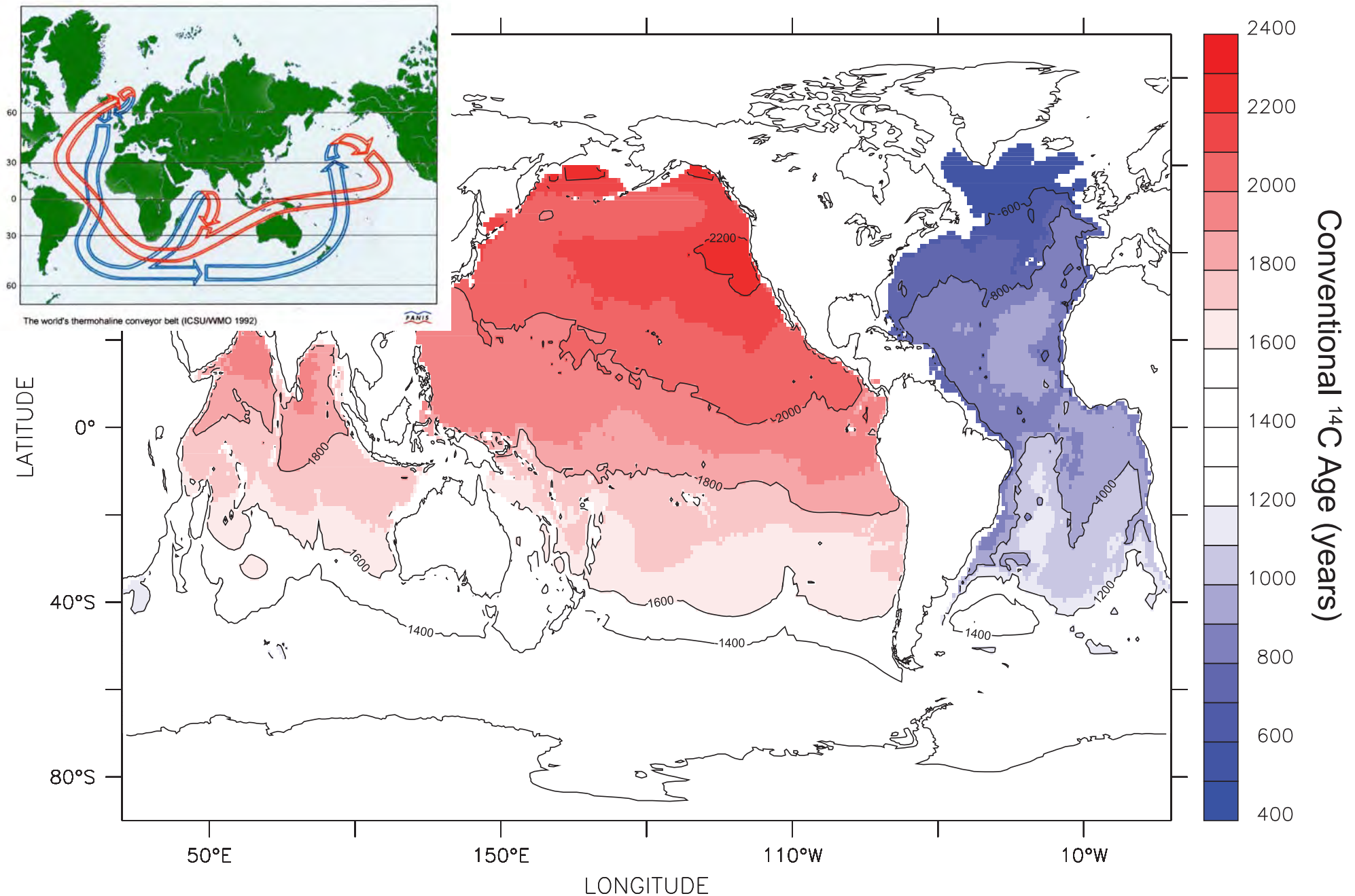
- Surface: ^{14}C exchange with atmosphere
age ~ 400 years
- Deep ocean: no ^{14}C exchange with atmosphere
age > 400 years, deep ocean can be very 'old'



The world's thermohaline conveyor belt (ICSU/WMO 1992)

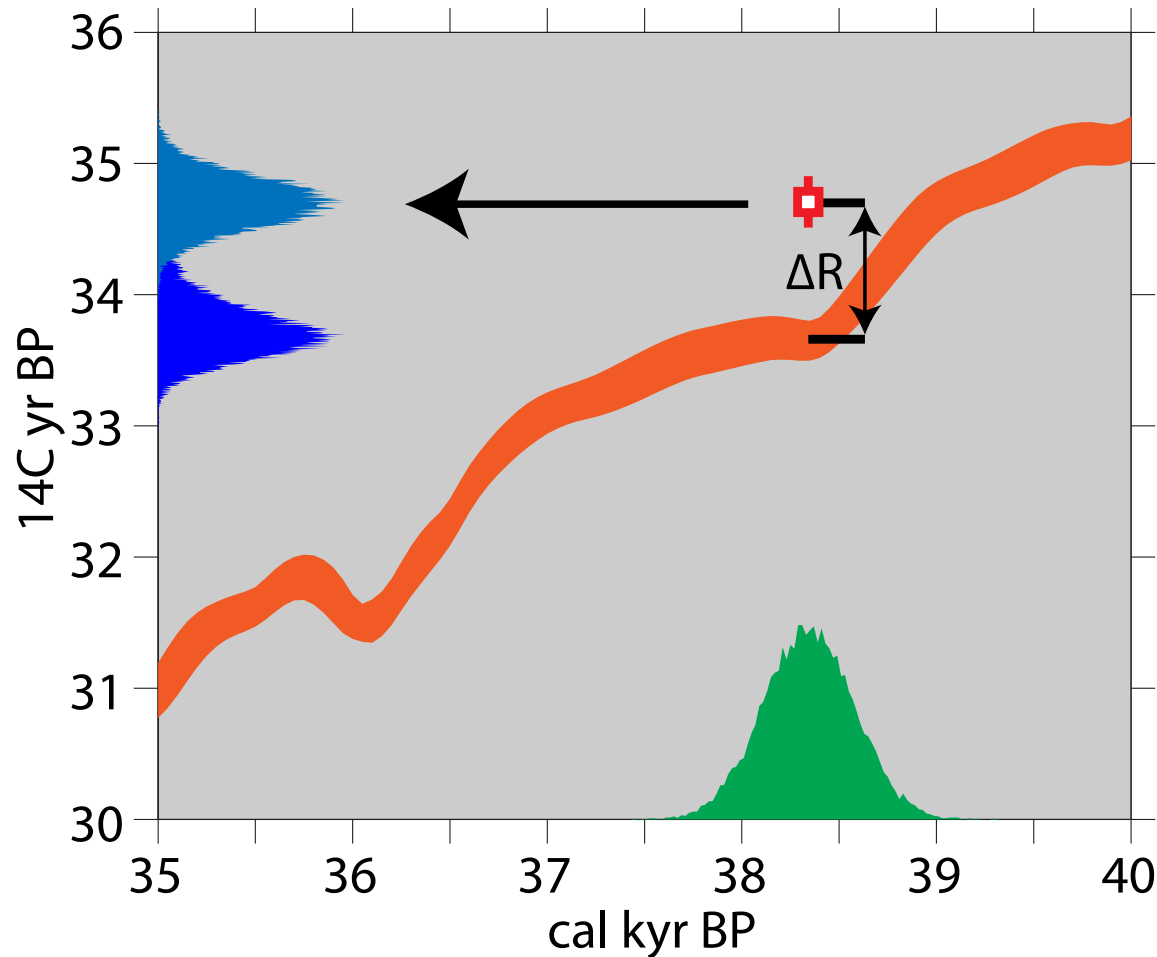


Deep ocean ^{14}C ages (>1500 m)



Matsumoto, K., 2007. *Journal of Geophysical Research-Oceans*, 112(C9): DOI: 10.1029/2007JC004095.

Calculating reservoir ages



Requires knowledge on **calendar** age
use 'atmospheric' ^{14}C samples
use tephro-chronology
use tie-points (*under the assumption of synchronicity*)

Reservoir age **R**
deviation from **atmospheric** curve

Local reservoir **ΔR**
deviation from **marine** curve

Marine chronology

www.radiocarbon.org/IntCal04

Global model ocean ~400 years older than contemporaneous atmosphere

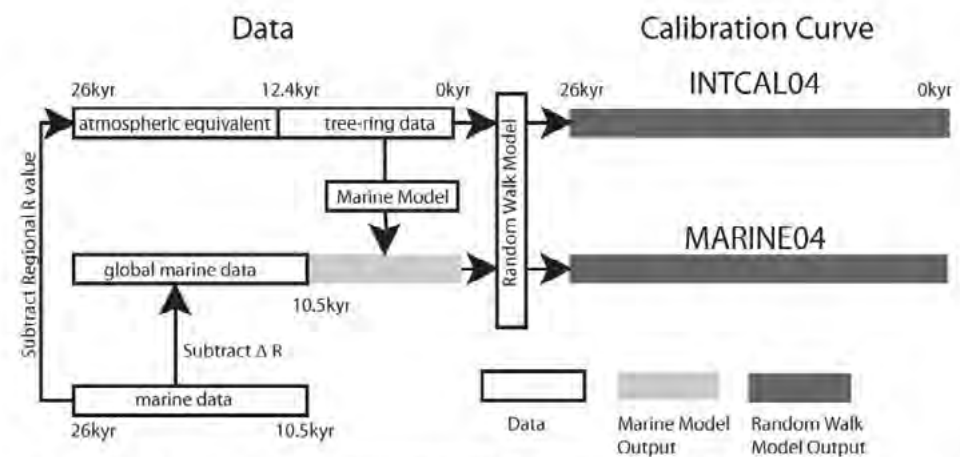
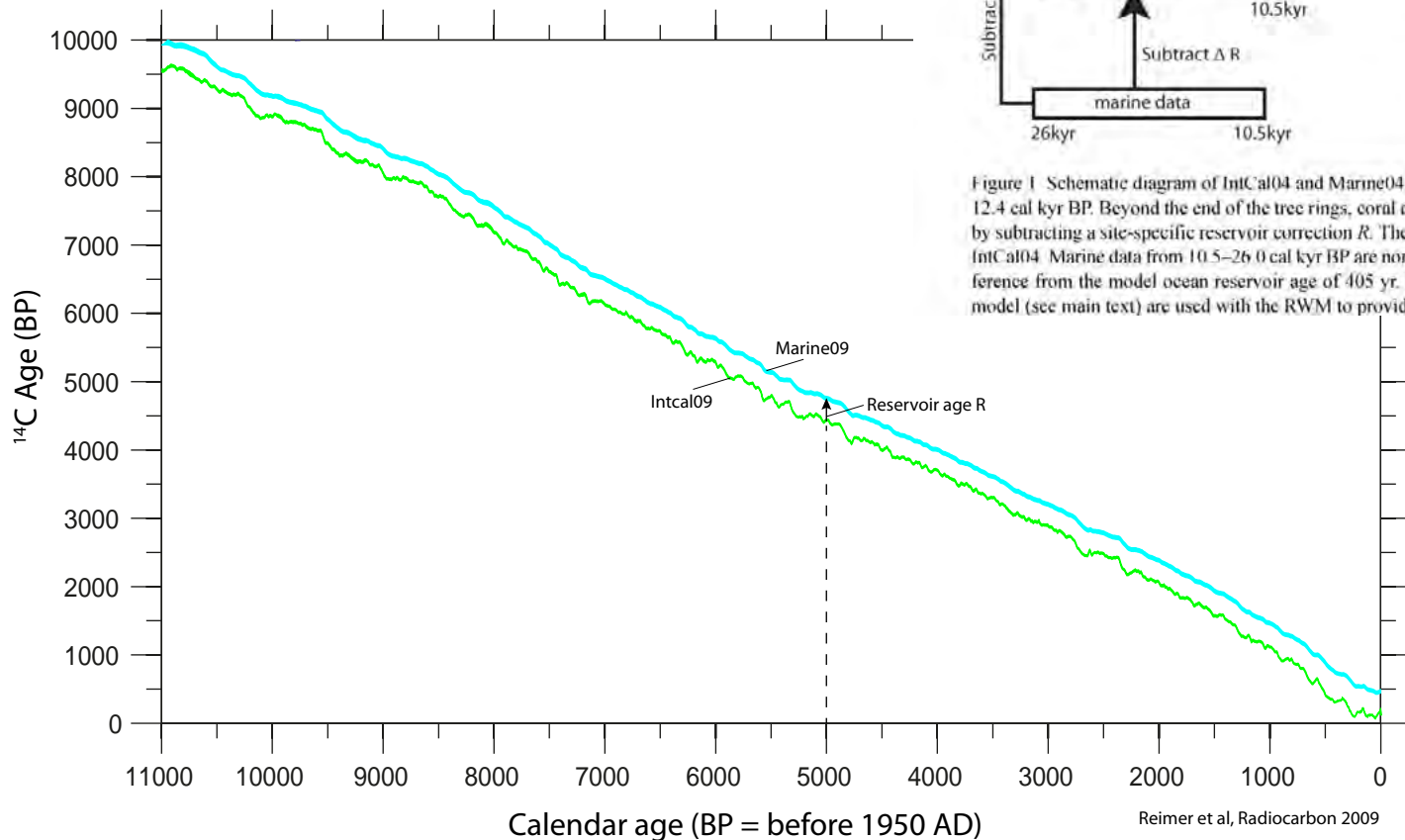
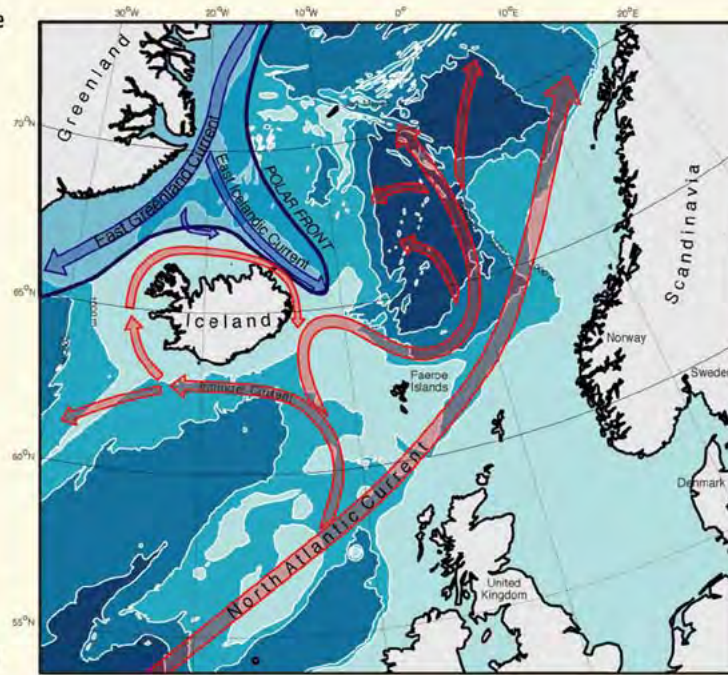


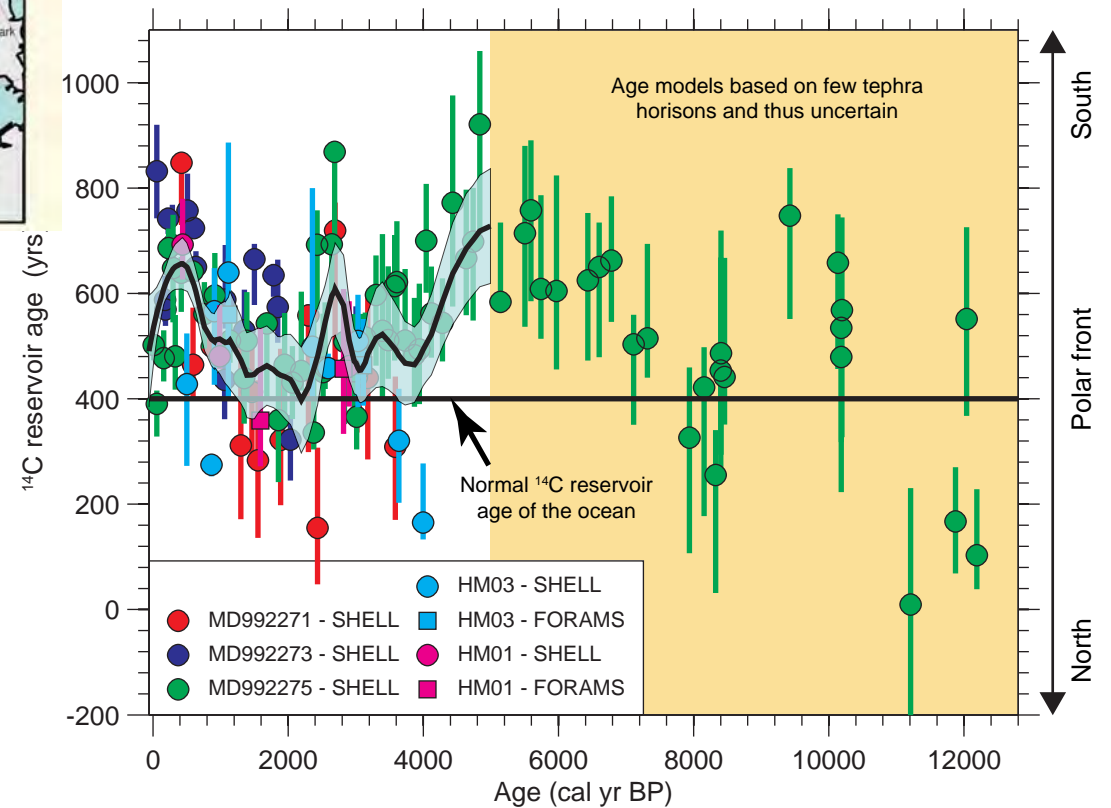
Figure 1 Schematic diagram of IntCal04 and Marine04 calibration data set construction. Tree-ring data extend from 0 to 12.4 cal kyr BP. Beyond the end of the tree rings, coral and foraminifera data are converted to the atmospheric equivalent by subtracting a site-specific reservoir correction R . These data are input into the random walk model (RWM) to produce IntCal04. Marine data from 10.5–26.0 cal kyr BP are normalized to the “global” ocean by subtracting ΔR , the regional difference from the model ocean reservoir age of 405 yr. The “global” marine data and the output from the box diffusion model (see main text) are used with the RWM to provide our estimate of the global marine calibration curve Marine04.

Modern Surface Circulation

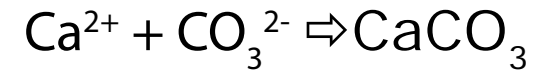
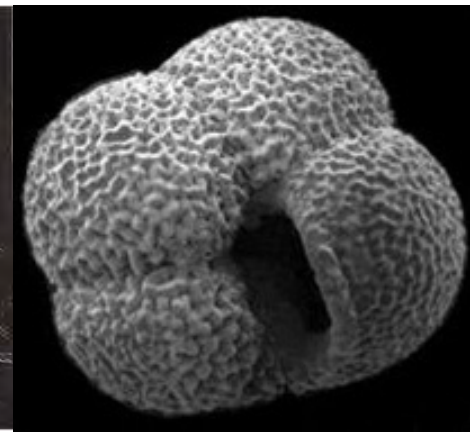


Modified after Map AB in The Nordic Seas (ed. G. Hurdle 1986)

Holocene ^{14}C variability linked to changes in position of polar front



How do we it?



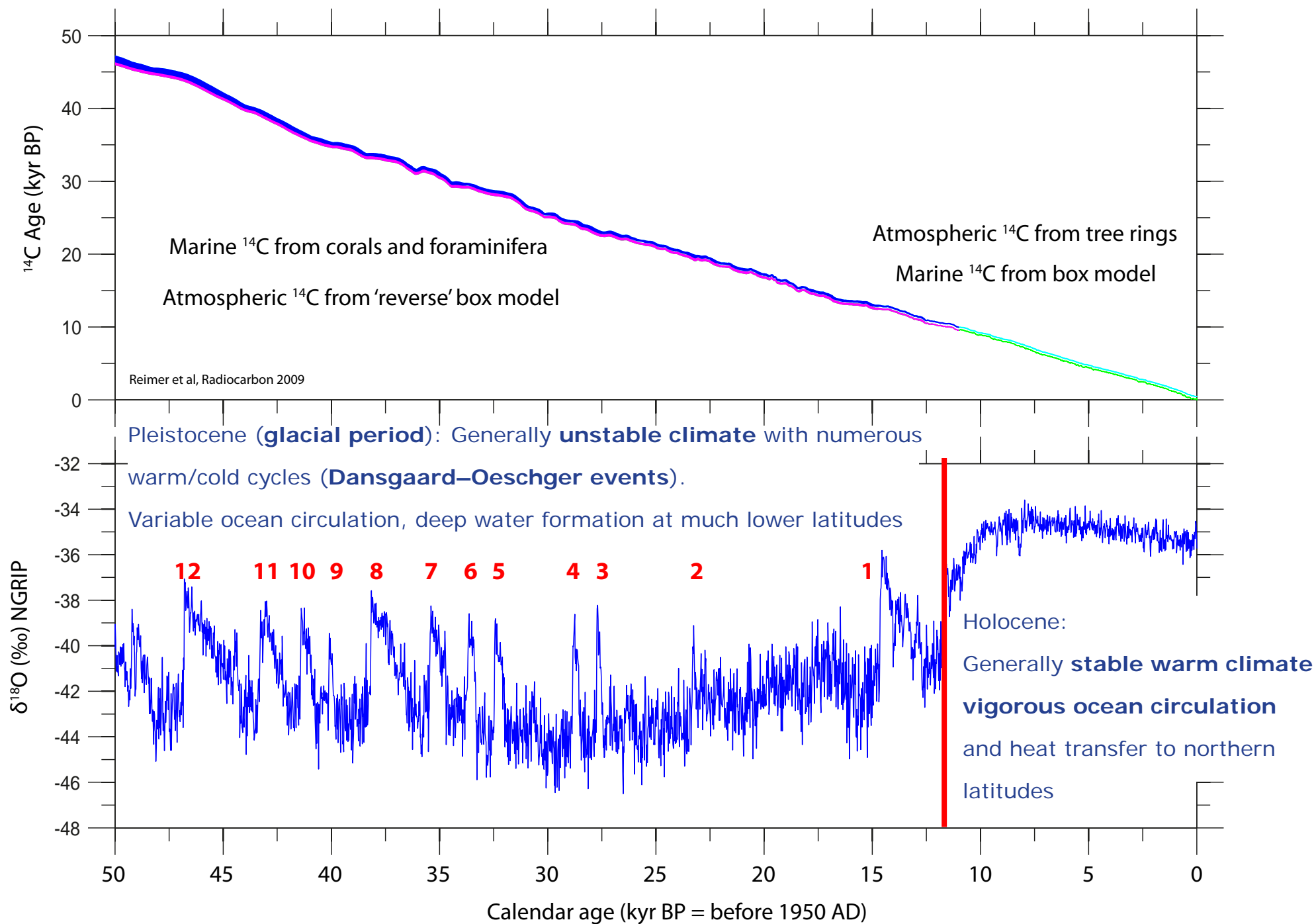
Shells and foraminifera calcify in equilibrium with dissolve ocean bicarbonate

proxy for ocean ^{14}C (**age**), $\delta^{13}\text{C}$ (**production / stratification**) and $\delta^{18}\text{O}$ (**salinity / temperature**)

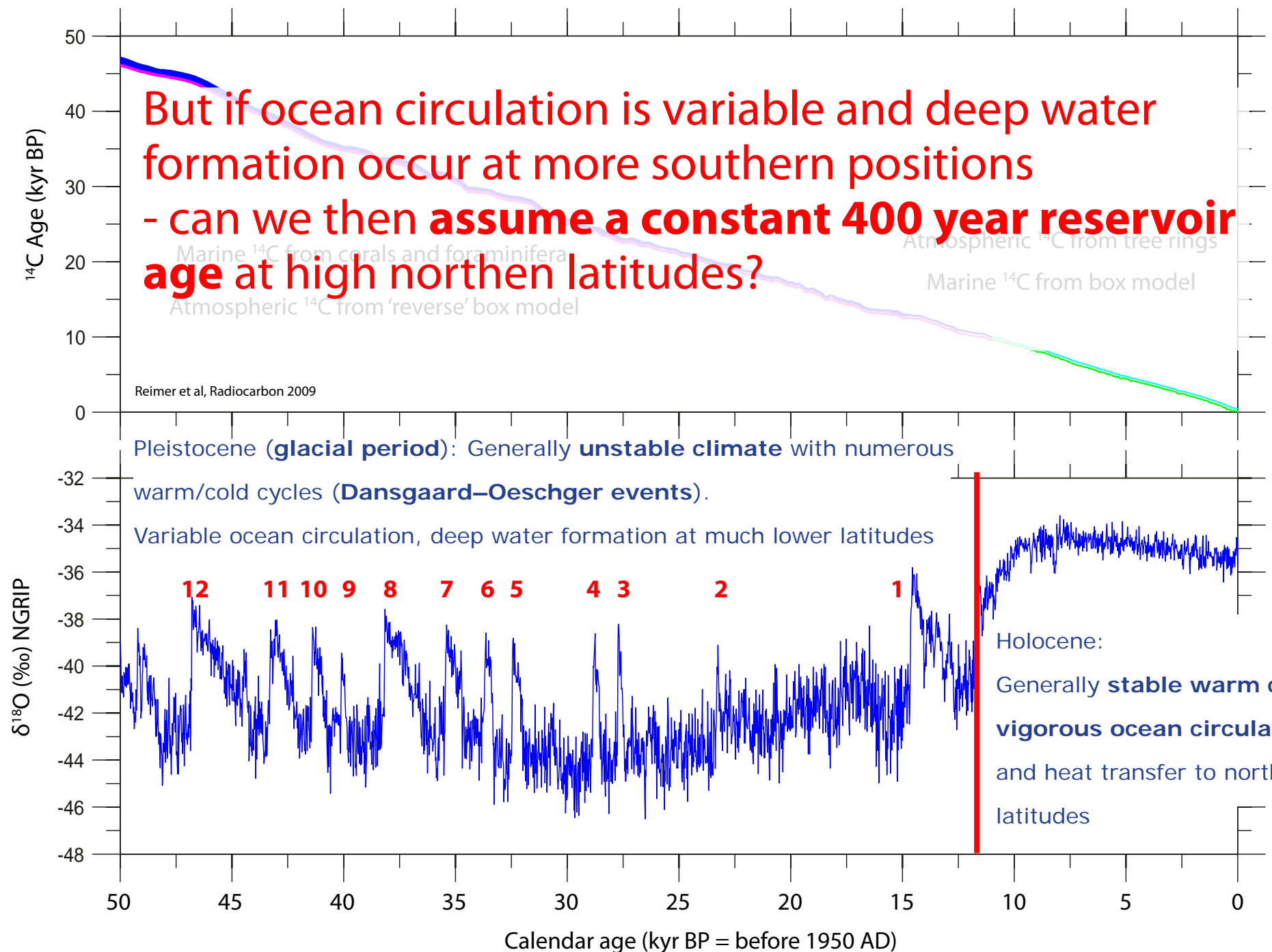
Planktic species provides information on **surface ocean** conditions (*100 - 200 metres*)

Benthic species provides information on **deep ocean** conditions (*few hundreds meters to kilometers*)

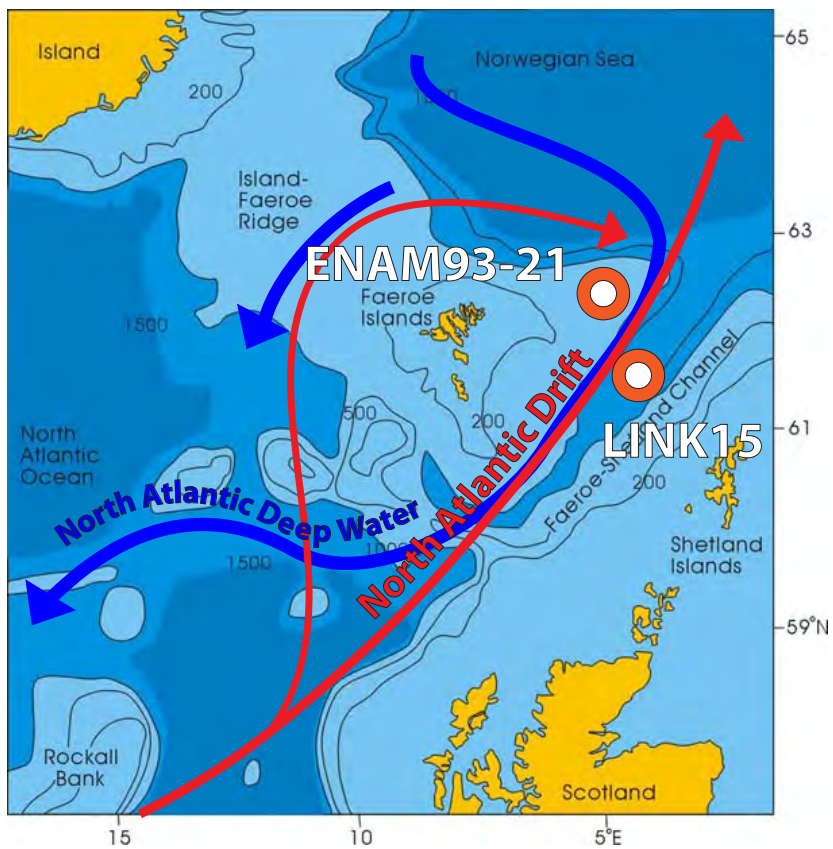
Going beyond the Holocene (last 11k)

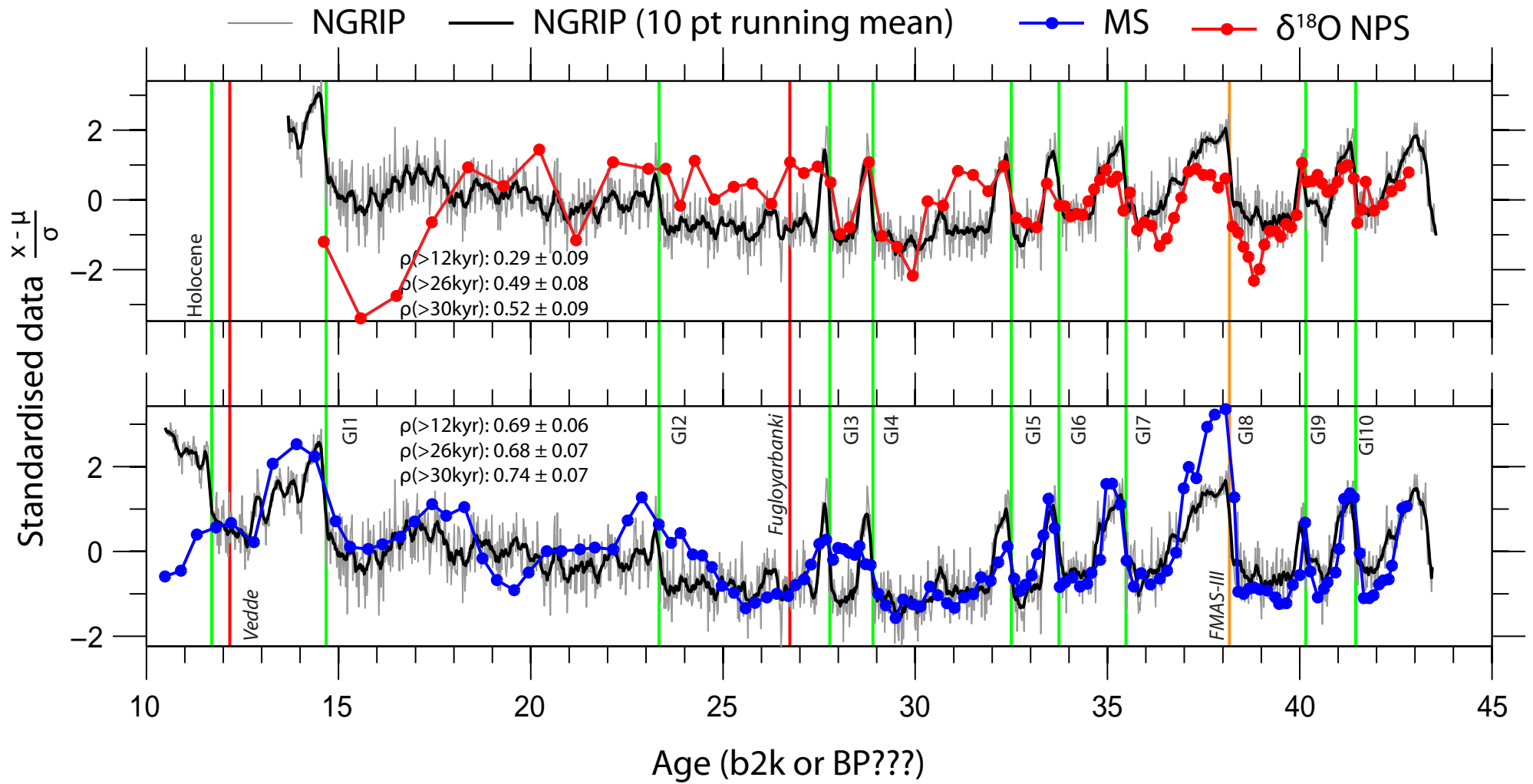


Going beyond the Holocene (last 11k)

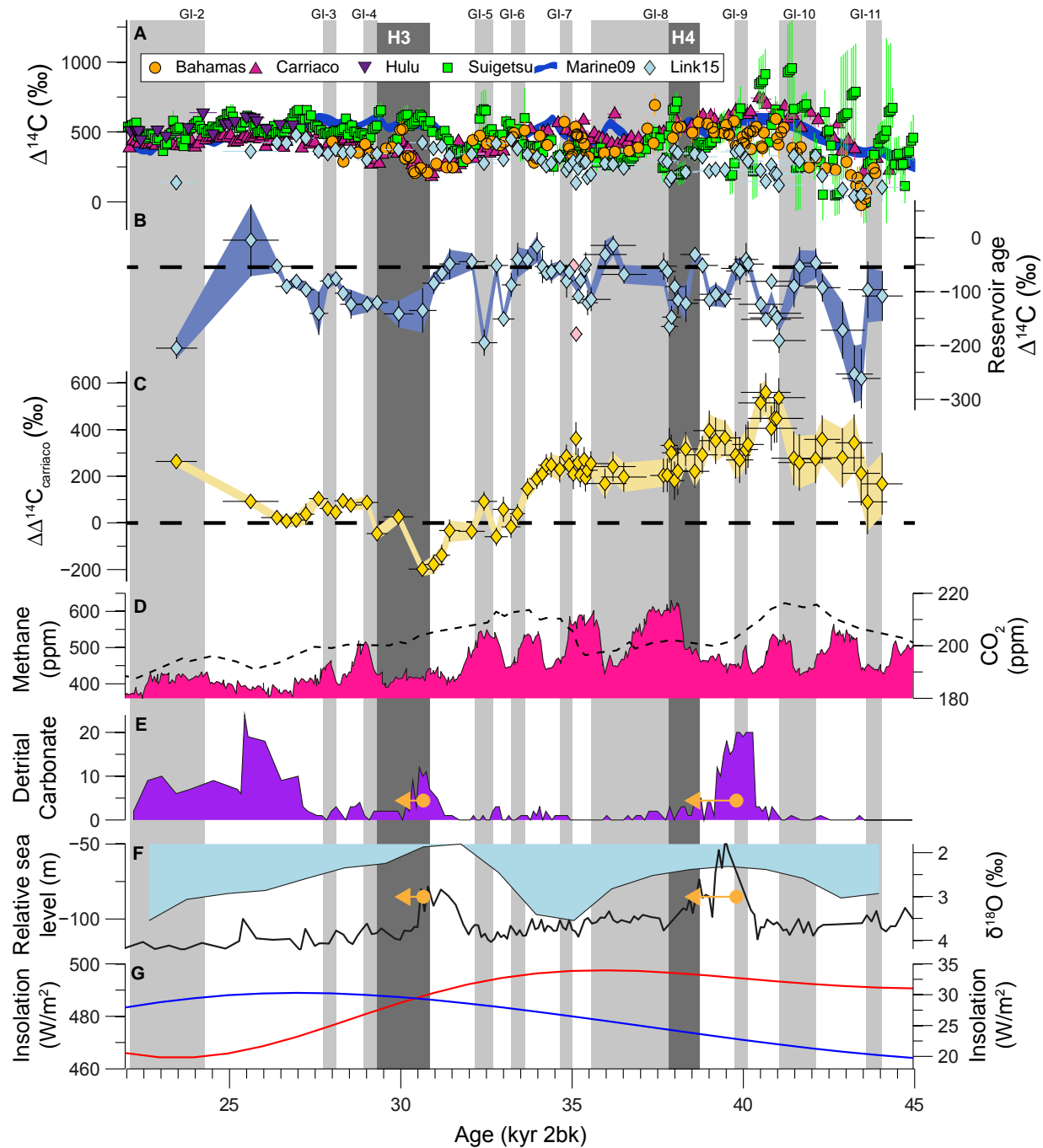


LINK15 - reservoir age variability 30 - 40 cal kyr BP

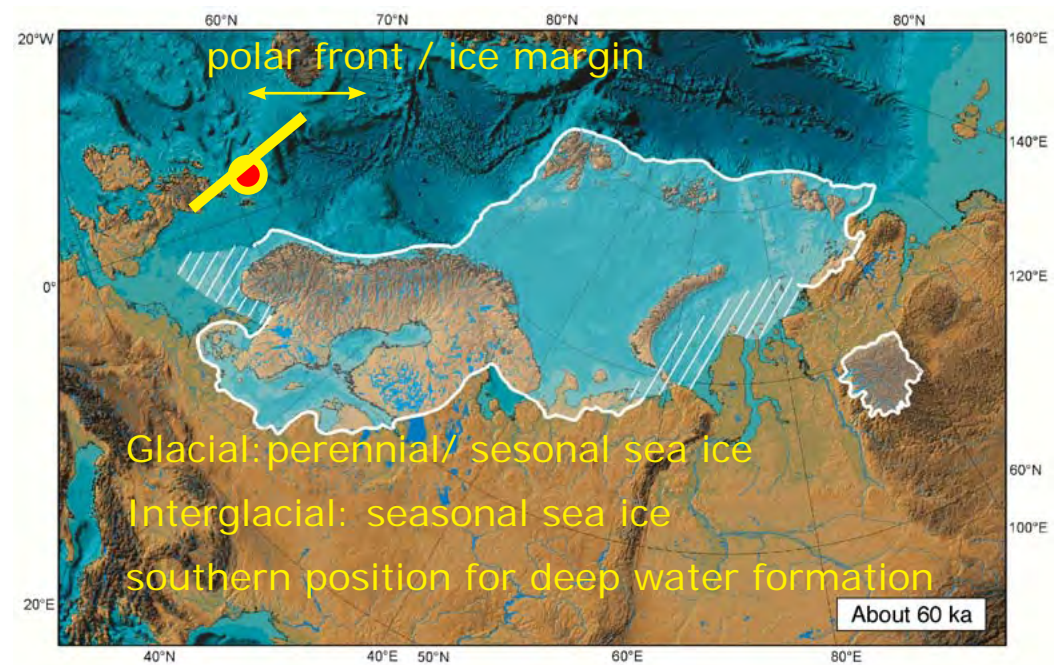
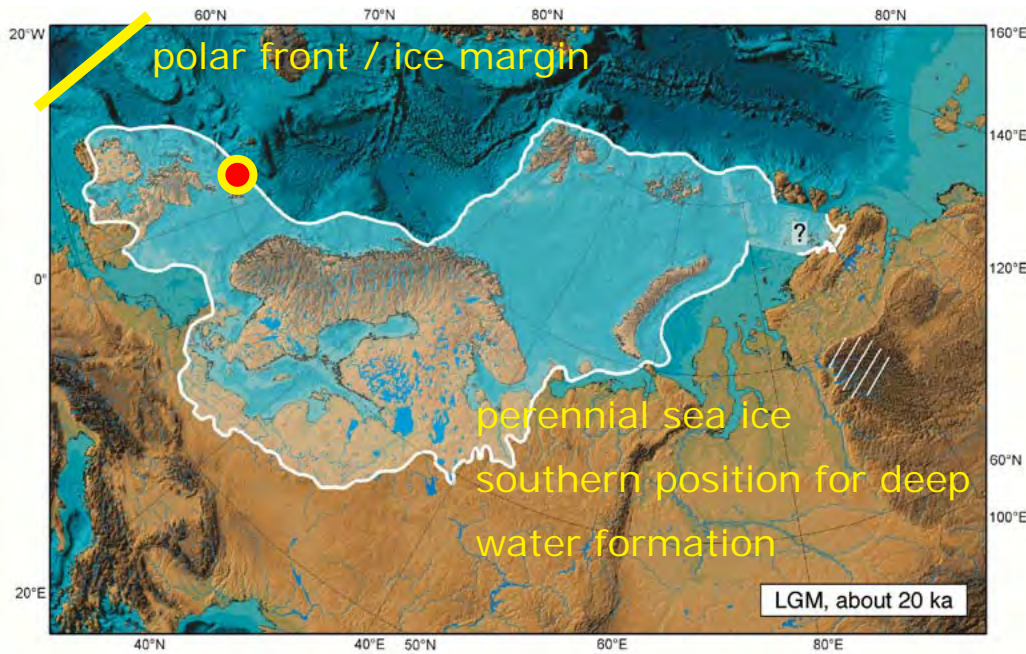




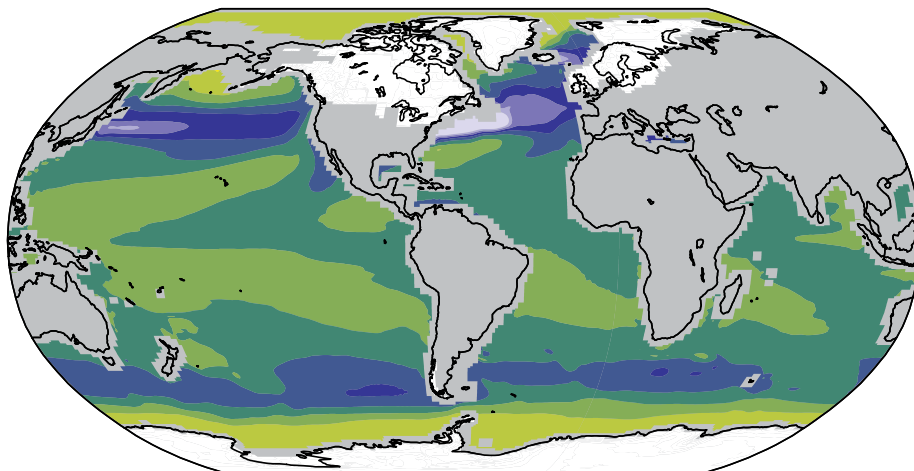
Significant change in R (up to R=2000 ^{14}C years)



Why the difference at LGM?



QRS 23 2004, APEX/ QUEENS



-7 -6 -5 -4 -3 -2 -1 0 1

Sea Surface Temperature Change (°C)

IPCC fig 6.7 (modified)

Position of the polar front associated with sea ice coverage

Tak for Jeres opmærksomhed

